

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 18. It exists in the periodic table.
- _____ 19. It has no line periodic table.
- _____ 20. Group 1 elements.
- _____ 21. Elements that are strong oxidizing agents.
- _____ 22. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 23. There are two representative elements of elements.
- _____ 24. Most elements have higher of the elements in the periodic table are metals.
- _____ 25. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 26. Group 2A elements are the alkaline earth metals.
- _____ 27. Group 1A elements are highly reactive elements known as alkali metals.
- _____ 28. Group 1A elements are very reactive elements known as representative elements.
- _____ 29. Alkali metals have properties of both metals and non-metals.

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Table of Contents Stars Study Guide For Content Mastery Answers

1. Understanding the eBook Stars Study Guide For Content Mastery Answers
 - The Rise of Digital Reading Stars Study Guide For Content Mastery Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Stars Study Guide For Content Mastery Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Stars Study Guide For Content Mastery Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Stars Study Guide For Content Mastery Answers
 - Personalized Recommendations
 - Stars Study Guide For Content Mastery Answers User Reviews and Ratings
 - Stars Study Guide For Content Mastery Answers and Bestseller Lists

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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Stars Study Guide For Content Mastery Answers
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Stars Study Guide For Content Mastery Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Stars Study Guide For Content Mastery Answers
 - Setting Reading Goals Stars Study Guide For Content Mastery Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Stars Study Guide For Content Mastery Answers
 - Fact-Checking eBook Content of Stars Study Guide For Content Mastery Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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