

C-100 Practice Multiple Choice Question Set 2

1. When the following reaction equation is properly balanced, the number of moles of O_2 will be...



- (a) 1.5 moles
- (b) 3.5 moles
- (c) 3 moles
- (d) 5 moles

2. When the following reaction equation is properly balanced, the number of moles of O_2 will be...



- (a) 1.5 moles
- (b) 13 moles
- (c) 19 moles
- (d) 38 moles

3. How many moles of CO_2 are present in 220 mg?

- (a) 5 moles
- (b) 0.005 mole
- (c) 5000 moles
- (d) 10 moles

4. If the temperature of a reaction is increased by $20^\circ C$, the reaction rate will be...

- (a) 2 times as fast.
- (b) 4 times as fast.
- (c) 20 times as fast.
- (d) unchanged because the reaction rate is not dependent on the temperature.

5. One mole of $(NH_4)_2HPO_4$ contains __?__ moles of hydrogen atoms.

- (a) 1
- (b) 5
- (c) 6
- (d) 9

6. Which of the following compounds should have the highest boiling point?

- (a) NH_3
- (b) CH_4
- (c) H_2O
- (d) HF

7. The following reaction coordinate diagram represents...



- (a) an endothermic reaction.
- (b) an exothermic reaction.

Static Equilibrium Multiple Choice Questions

S Ben Porath



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