

Starfish Dissection

Procedure 2: Self-Guided Starfish Dissection

Materials: Preserved starfish, dissecting pan, scissors, scalpel, forceps, T-pins, pencil, lab apron, safety glasses

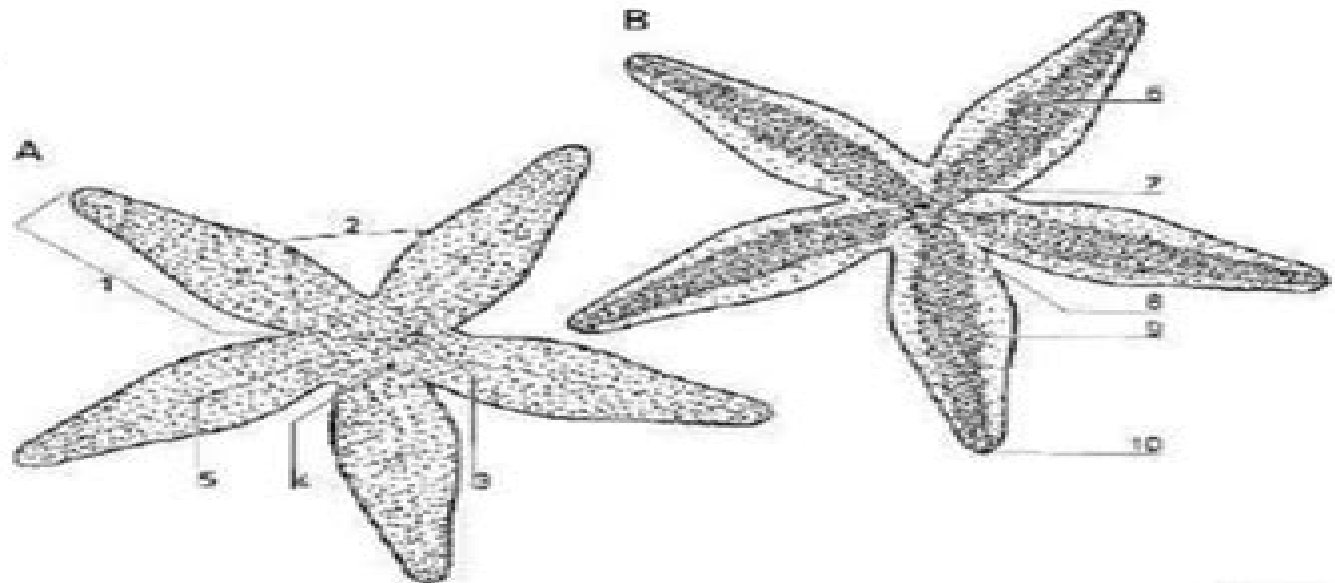
Procedure (Aboral Surface):

1. Obtain a preserved starfish and rinse off preservative with tap water.
2. Place the starfish in the dissecting pan with its dorsal or aboral (top) surface upward.
3. Observe the starfish and determine its **symmetry**. _____
4. Locate the **central disc**. Count and record the number of arms or rays the starfish has. _____
5. Locate the small, round hard plate called the **madreporite** on top of the central disc. Water enters through this into the water vascular system. Label the **central disc**, **arms**, and **madreporite** on **Figure 1A**.
6. The **anus**, for excretion following digestion, is a smaller opening right next to the **madreporite**. Label this on **Figure 1A**.
7. Feel the upper surface of the starfish for **spines**. These spines protect the starfish and are part of their **internal skeleton**. Label this on **Figure 1A**. Between each spine are **skin gills**. These are pores that allow the starfish to take in oxygen for cellular respiration.

Figure 1: A = Aboral Surface, B = Oral Surface

Procedure (Oral Surface):

8. Turn the starfish over to its ventral or oral surface (underside).
9. Look at the tip of each arm and find the **eyespot**. Label this on **Figure 1B**.
10. Locate the **mouth** in the center of the central disc. Find the ring of **oral spines** surrounding the mouth. Label these structures on **Figure 1B**.
11. Find the groove that extends down the underside of each arm. This is called the **ambulacral groove**. Label this on **Figure 1B**.
12. Feel the numerous, soft **tube feet** inside each groove. These are part of the **water vascular system** & aid in movement and feeding. Label these on **Figure 1B**.



Starfish Dissection Questions Answers

Buel Preston Colton



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The St. Andrews University Calendar for the Year ... University of St. Andrews, 1900 Science Examinations ... Reports, Etc, 1909 *Reconceptualizing the Literacies in Adolescents' Lives* Donna E. Alvermann, Kathleen A. Hinchman, David W. Moore, Stephen F. Phelps, Diane R. Waff, 2007-07-10 Reconceptualizing the Literacies in Adolescents Lives Second Edition focuses on exploring the impact of young people s identity making practices in mediating their perceptions of themselves as readers and writers in an era of externally mandated reforms What is different in the Second Edition is its emphasis on the importance of valuing adolescents perspectives in an era of skyrocketing interest in improving literacy instruction at the middle and high school levels driven by externally mandated reforms and accountability measures A central concern is the degree to which this new interest takes into account adolescents personal social and cultural

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High School Department, 1903 *American Medical Association Bulletin* American Medical Association, 1912 **Bulletin**
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web allow the gaseous phase thermodynamics to be curve fitted to 300k about 142k lower than the normal boiling point 542k 2 of kerosene nevertheless in actual calculations involving liquid kerosene fuel the heat capacity and latent heat of vaporization of liquid kerosene 12 are used for

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web as a general trend vapor pressures of liquids at ambient temperatures increase with decreasing boiling points this is illustrated in the vapor pressure chart see right that shows graphs of the vapor pressures versus temperatures for a variety of liquids

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web the vapor pressure of a liquid is defined as the pressure exerted by the molecules that escapes from the liquid to form a separate vapor phase above the liquid surface the pressure exerted by the vapor phase is called the vapor or saturation pressure vapor or saturation pressure depends on temperature

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web sep 21 2022 a vapor pressure curve is a graph of vapor pressure as a function of temperature to find the normal boiling point of liquid a horizontal line is drawn from the y axis at a pressure equal to standard pressure

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web the graph of the vapor pressure of water versus temperature in figure page index 3 indicates that the vapor pressure of

water is 68 kpa at about 90 c thus at about 90 c the vapor pressure of water will equal the atmospheric pressure in

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web vapor pressure 0.1 psi at 100 f ntp 1992 vapor density relative to air 4.5 ntp 1992 heavier than air will sink specific gravity 0.8 at 59 f uscg 1999 less dense than water will float

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web in fact when the temperature increases from 20 to 60 c the vapor pressure of kerosene rises from 4 to 18 mbar 34 this suggests that kerosene cannot be entirely vaporized at 60 c

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web kerosene icsc 0663 physical chemical information physical state appearance low viscosity liquid with characteristic odour physical dangers as a result of flow agitation etc electrostatic charges can be generated vapour pressure kpa at 37.8 c 1.37 exposure health effects