

Standardized Test prep Answers

Chapter 1 Page 25

1. C 4. B 7. C
2. C 5. C 8. A
3. D 6. C 9. B

10. Answers may vary.

11. Observations C and G are chemical properties; the others are physical properties.

12. In a chemical change, one or more substances are converted into different substances. A physical change does not involve a change in the identity of the substance or substances present.

13. metals: shiny; good conductors of heat; good conductors of electricity; malleable or ductile; most are solids at room temperature
nonmetals: poor conductors of heat; poor conductors of electricity; many are gases at room temperature; those that are solids are brittle rather than malleable or ductile metalloids: properties intermediate between those of metals and nonmetals; less malleable than metals but not as brittle as solid nonmetals; most are semiconductors of electricity

Chapter 3 Page 93

1. C 4. A 7. B
2. D 5. B 8. D
3. C 6. A 9. C

10. Argon-40 has 22 neutrons ($40 - 18 = 22$), and potassium-40 has 21 neutrons ($40 - 19 = 21$).

11. 6.17 g

12. All cathode rays are the same, regardless of their source. Therefore, the particles responsible for the cathode rays must be present in all atoms. The particles are electrons.

13. When the average atomic mass is calculated, it is 10.811. Because the atomic mass is the same as the atomic mass of boron, mythium was not a new element.

Chapter 5 Page 1

1. C
2. A
3. A
4. A
5. C
6. A
7. D
8. B
9. C

10. S: 2251 kJ/mol; Cl: 2297 kJ/mol; Ar: 2666 kJ/mol; K: 3051 kJ/mol; Ca: 1345 kJ/mol; Sc: 1235 kJ/mol; Ti: 1310 kJ/mol. For the second ionization, the general trend is for increasing IE2 across the period in Groups 2–18 with Group 1 having the highest IE2. IE2 decreases going down a group.

11. Group 16 most commonly forms 2- ions, because these elements require only two more electrons to fill their shell (obtain a noble-gas configuration).

Chapter 2 Page 63

1. B 4. B 7. A
2. C 5. A 8. B
3. C 6. B 9. B

10. 2 cups; direct proportion

11. Statements A and B contain exact numbers.

12. The type of fertilizer is the variable being tested. Control factors are the types of radishes, the amount of water and the amount of sunshine. One control row should be planted under the same control factors but with no fertilizer. There are at least four things that could be used to determine the results: size, quantity, appearance, and taste. Analysis might include bar graphs of each of these measurements for each of the five fertilizer types and the no-fertilizer control row.

13. A unit must be defined in a way that does not depend on the circumstances of the measurement. Not every thumbnail is the same size.

Chap 4 Page 129

1. A 4. D 7. C
2. D 5. B 8. B
3. A 6. B 9. C

10. The color will be orange. Converting energy into frequency gives 4.8×10^{14} , which corresponds to the frequency of orange light.

11. $\frac{1}{3s}$ $\frac{1}{3s}$ $\frac{1}{3p}$ $\frac{1}{3p}$

12. Electrons in atoms can occupy orbitals of only specific energies. When an atom is excited, the electron is no longer in the ground state. When the electron returns to a lower energy level, light is emitted. Because only specific energies are allowed, certain wavelengths of light are emitted, giving rise to the individual lines in the spectrum.

13. Photons of blue light are higher energy than photons of yellow light. Electrons can be emitted only when a photon of sufficient energy strikes the surface of the metal. Therefore, the energy of blue light is greater than the threshold energy, but the energy of yellow light is not.

Chapter 6 Page 215

1. A 2. C 7. A
3. B 4. D 8. B
5. A 6. D 9. C

10. Hybridization explains how the orbitals of an atom become rearranged when the atom forms covalent bonds.

11. Ionic crystals are brittle because shifting of the layers of ions results in large repulsive forces that cause the layers to part completely.

12. Naphthalene has the stronger intermolecular forces even though it is non-polar, because its boiling point is higher than that of acetic acid. Boiling point is directly correlated to strength of intermolecular forces; the stronger the intermolecular forces, the more energy needed to break all the intermolecular forces, and therefore the higher the boiling point. Naphthalene is so large that its dispersion forces are greater than the sum of the dispersion forces and hydrogen bonding in acetic acid.

13. There is a repulsive region at small bond distances, a minimum well at an intermediate bond distance, and an asymptotic region at large bond distances. The repulsive region is due to the repulsion between the two positive nuclei at distances where the nuclei are close to each other. The asymptotic region is where the atoms are far apart and therefore do not interact with each other. The potential energy decreases as the atoms come closer together and are attracted to each other. The minimum well is the region where the attractive forces are balanced by the repulsive forces. This distance is the equilibrium bond length.

Usa Test Prep Chemistry Answer Key

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Army**



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ABOUT THE TEST The Advanced Placement Program is designed to allow high school students to pursue college level studies while attending high school The three hour five minute AP U S History exam is usually given to high school students who have completed a year s study in a college level U S History course The test results are then used to determine the awarding of course credit and or advanced course placement in college According to the College Board students taking this exam are called upon to demonstrate systematic factual knowledge and bring to bear critical persuasive analysis of the full sweep of U S history This is why we make every effort to establish and build upon context for you rather than encouraging rote memorization of disconnected facts The AP U S History Exam is divided into two sections

1 Multiple Choice This section is composed of 80 multiple choice questions designed to gauge your ability to understand and analyze U S history from the Colonial period to the present The majority of the questions however are based on 19th and 20th century history This section tests factual knowledge scope of preparation and knowledge based analytical skills You ll have 55 minutes to complete this section which accounts for 50 percent of your final grade

2 Free Response This section is composed of three essay questions designed to measure your ability to write coherent intelligent well organized essays on historical topics The essays require you to demonstrate mastery of historical interpretation and the ability to express views and knowledge in writing The essays may relate documents to different areas analyze common themes of different time periods or compare individual and group experiences which reflect socioeconomic racial gender and ethnic differences

Part A consists of a mandatory 15 minute reading period followed by 45 minutes during which you must answer a document based question DBQ which changes from year to year In **Part B** the student chooses to answer on two of the topics that are given You will have 70 minutes to write these essays The free response section counts for 50 percent of your final grade These topics are broken down into thirds

Political Institutions 1 3rd **Social and Economic Change** 1 3rd **Behavior and Public Policy** **Diplomacy and International Relations** **Intellectual and Cultural Development** 1 3rd

The time periods covered are as follows

Pre Colonial through 1789 1 6th of exam **1790 1914** 1 2 of exam **1915 present** 1 3rd of exam

ABOUT THE REVIEW SECTION This book begins with REA s concise yet thorough 230 page review of U S history designed to acquaint you with the exam s scope of coverage Our review covers these topics and historical time periods

The Colonial Period 1500

1763 The American Revolution 1763 1787 The United States Constitution 1787 1789 The New National 1789 1824 Jacksonian Democracy and Westward Expansion 1824 1850 Sectional Conflict and The Causes of the Civil War 1850 1860 The Civil War and Reconstruction 1860 1877 Industrialism War and the Progressive Era 1877 1912 Wilson and World War I 1912 1920 The Roaring Twenties and Economic Collapse 1920 1929 The Great Depression and the New Deal 1929 1941 World War II and the Post War Era 1941 1960 The New Frontier Vietnam and Social Upheaval 1960 1972 Watergate Carter and the New Conservatism 1972 2001

SCORING THE EXAM The multiple choice section of the exam is scored by crediting each correct answer with one point and deducting one fourth of a point for each incorrect answer You will neither receive a credit nor suffer a deduction for unanswered questions The free response essays are graded by instructors and professors from across the country who come together each June for a week of non stop AP essay grading Each essay booklet is read and scored by several graders Each grader provides a score for the individual essays The DBQ is scored on a scale from 0 to 15 0 being the lowest and 15 the highest Each topic based essay receives a score from 0 to 9 These scores are concealed so that each grader is unaware of the previous graders assessments When the essays have been graded completely the scores are averaged one score for each essay so that the free response section generates three scores The total weight of the free response section is 50 percent of the total score Your work in the multiple choice section counts for the other 50 percent Each year grades fluctuate slightly because the grading scale is adjusted to take into account the performance of the total AP U S History test taker population When used with the corresponding chart the scoring method we present here will strongly approximate the score you would receive if you were sitting for the actual AP U S History exam

SCORING THE MULTIPLE CHOICE SECTION For the multiple choice section use this formula to calculate your raw score Number right number wrong x 1 4 raw score round to the nearest whole number

SCORING THE FREE RESPONSE SECTION For the free response section use this formula to calculate your raw score DBQ Essay 1 Essay 2 raw score round to the nearest whole number You may want to give your essays three different grades such as a 13 10 and an 8 and then calculate your score three ways as if you di

Pamphlet - Dept. of the Army United States. Department of the Army,1951 **The United States Catalog** ,1921

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Clarke, Mark Marzinke, 2020-06-11 Contemporary Practice in Clinical Chemistry Fourth Edition provides a clear and concise overview of important topics in the field This new edition is useful for students residents and fellows in clinical chemistry and pathology presenting an introduction and overview of the field to assist readers as they in review and prepare for board certification examinations For new medical technologists the book provides context for understanding the clinical utility of tests that they perform or use in other areas in the clinical laboratory For experienced laboratorians this revision continues to provide an opportunity for exposure to more recent trends and developments in clinical chemistry Includes enhanced illustration and new and revised color figures Provides improved self assessment questions and end of chapter assessment questions The United States Catalog ,1918 **Energy and water development appropriations for fiscal year 1984**

United States. Congress. House. Committee on Appropriations. Subcommittee on Energy and Water Development, United States. Congress. Senate. Committee on Appropriations. Subcommittee on Energy and Water Development, 1983 **The Best Test Preparation for the Advanced Placement Examination in Mathematics, Calculus AB** Donald E. Brook, 1992

SAT II Gary Land, Ronald John Lettieri, 1990-01-01 REA s test preparation book for the SAT II United States History includes six full length practice exams based on official exam questions released by the College Board Also included is a comprehensive review course of US History covering all major topics found on the exam For college bound high school students who want the best test preparation for this exam Energy and water development appropriations for fiscal year 1984 United States. Congress. Senate. Committee on Appropriations. Subcommittee on Energy and Water Development, 1983

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