

Chemistry Spring Final Study Guide

Unit 10: Acids & Bases

Use the following information and complete the structure of Acids & Bases.

Acids	Bases
Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A solution of the perfect... 2. A solution of... 3. A solution of... 4. The gas is... 5. The solution...	Unit 10: Acids & Bases Draw a line on the graph and indicate its dependence for each property: Temperature, Volume, and Pressure below. Label the y-axis. Label the x-axis. 1. A solution of the perfect... 2. A solution of... 3. A solution of... 4. The gas is... 5. The solution...

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Name: _____

Unit 1: Measurement Analysis

Mark the best answer from those provided to each scenario.

Mark the best answer from those provided to each scenario.

Determine the number of Significant Figures in each of the following numbers.

1. 123.456 $\rightarrow$ _____ sig figs

2. 123.456 $\rightarrow$ _____ sig figs

3. 123.456 $\rightarrow$ _____ sig figs

4. 123.456 $\rightarrow$ _____ sig figs

Convert the following into Scientific Notation.

1. 123.456 $\rightarrow$ _____

2. 123.456 $\rightarrow$ _____

3. 123.456 $\rightarrow$ _____

4. 123.456 $\rightarrow$ _____

Convert the following into Decimal Notation.

1. 1.23×10^2 $\rightarrow$ _____

2. 1.23×10^2 $\rightarrow$ _____

3. 1.23×10^2 $\rightarrow$ _____

4. 1.23×10^2 $\rightarrow$ _____

Unit 2: Chemical Reactions

Mark the best answer from those provided to each scenario.

Mark the best answer from those provided to each scenario.

Determine the number of Significant Figures in each of the following numbers.

1. 123.456 $\rightarrow$ _____ sig figs

2. 123.456 $\rightarrow$ _____ sig figs

3. 123.456 $\rightarrow$ _____ sig figs

4. 123.456 $\rightarrow$ _____ sig figs

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2. 1.23×10^2 $\rightarrow$ _____

3. 1.23×10^2 $\rightarrow$ _____

4. 1.23×10^2 $\rightarrow$ _____

Unit 3: Reaction

For the following compounds, calculate the Molar Mass and the Percent Composition of each element.

H_2O

1. 18.02 g/mol

2. 18.02 g/mol

3. 18.02 g/mol

4. 18.02 g/mol

$NaCl$

1. 58.44 g/mol

2. 58.44 g/mol

3. 58.44 g/mol

4. 58.44 g/mol

$NaOH$

1. 40.00 g/mol

2. 40.00 g/mol

3. 40.00 g/mol

4. 40.00 g/mol

Na_2CO_3

1. 106.01 g/mol

2. 106.01 g/mol

3. 106.01 g/mol

4. 106.01 g/mol

Unit 4: Stoichiometry

Mark the best answer from those provided to each scenario.

Mark the best answer from those provided to each scenario.

Determine the number of Significant Figures in each of the following numbers.

1. 123.456 $\rightarrow$ _____ sig figs

2. 123.456 $\rightarrow$ _____ sig figs

3. 123.456 $\rightarrow$ _____ sig figs

4. 123.456 $\rightarrow$ _____ sig figs

Convert the following into Scientific Notation.

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3. 123.456 $\rightarrow$ _____

4. 123.456 $\rightarrow$ _____

Convert the following into Decimal Notation.

1. 1.23×10^2 $\rightarrow$ _____

2. 1.23×10^2 $\rightarrow$ _____

3. 1.23×10^2 $\rightarrow$ _____

4. 1.23×10^2 $\rightarrow$ _____

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Rachel Sandford



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Bulletin of Information United States Coast Guard Academy, Broadening Participation in STEM Zayika Wilson-Kennedy, Goldie S. Byrd, Eugene Kennedy, Henry T. Frierson, 2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines *Research in Education*, 1974 University Record University of Chicago, 1903

Undergraduate Catalog ... with Announcements for ... Western Michigan University, 1929 *University Record*, 1903

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Basics of Ecotoxicology Donald W. Sparling, 2017-07-28 This textbook presents a comprehensive examination of environmental science and ecotoxicology for undergraduate students The material provides sufficient related background information leading to a competency to clearly understand ecotoxicology concepts and topics

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