

PAPER II

Answer any **four** questions only.

1. a. (i) What is fermentation? (2 marks)
 (ii) With chemical equations only, show how ethanol can be produced from starch? (6 marks)

b. Complete Table 2.1.

Table 2.1: Characteristics of halogens

Halogen	State at room temperature	Uses
Fluorine	(1 mark)	Manufacture of teflon
Chlorine	(1 mark)	Manufacture of antiseptics

- c. (i) How many moles are contained in 25 g of magnesium trioxocarbonate (IV)? (2 marks)
 [Mg = 24, C = 12, O = 16] (3 marks)
- (ii) Name a suitable indicator for the titration of strong acid against weak base. (1 mark)
- (iii) Give an example of:
 I. Acidic oxide
 II. Neutral oxide (2 marks)
- (iv) Give **one** difference between hydrophobic and hydrophilic compounds. (2 marks)
- (v) Write the IUPAC name of the following compounds:
 I. $\text{CH}_3\text{CH}_2\text{COOH}$
 II. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ (2 marks)
- d. (i) Define the rate of chemical reaction. (2 marks)
 (ii) State **three** factors that affect the rate of chemical reaction. (3 marks)

Weac2014 Chemistry Essay And Objective Answer

Christian Drosten



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