#### Reading a Solubility Chart

- The curve shows the # of grams of solute in a saturated solution containing 100 mL or 100 g of water at a certain temperature. (remember: 1 mL H<sub>2</sub>O = 1 g H<sub>2</sub>O)
- Any amount of solute below the line indicates an unsaturated solution at a certain temperature
- Any amount of solute above the line in which all of the solute has dissolved indicates a supersaturated solution.
- 4) If the amount of solute is above the line but has not all dissolved, the solution is saturated and the # grams of solute settled on the bottom of the container = total # g in solution - # g of a saturated solution at that temperature. (according to the curve)
- Solutes whose curves move upward w/ increased temperature are typically solids b/c the solubility of solids increases w/ increased temperature.
- 6) Solutes whose curves move downward w/ increased temperature are typically gases b/c the solubility of gases decreases with increased temperature.



#### Solubility Problems example (Many may be done by reading the solubility table)

- At 10°C, 80 g of NaNO<sub>3</sub> will dissolve in 100 mL (a saturated solution)
- To find the # grams needed to saturate a solution when the volume is NOT 100 mL use the following strategy to find answer: (e. g., 60 mL of H<sub>2</sub>O will be saturated with how many grams of NaNO<sub>3</sub>?)

Start w/ known vol. x Solubility/100mL at set temp. = amount of Solute needed to saturate

Ex. 60 mL H<sub>2</sub>O x 80 g NaNO<sub>3</sub> = 48 g NaNO<sub>3</sub> needed to saturate solution 100 mL H<sub>2</sub>O

or if the chart is in units of 100 g of H2O use the density of water conversion 1mL H2O= 1 g H2O

Ex.  $60 \text{ mL H}_2\text{O} \text{ x} \quad \frac{1 \text{ g H}_2\text{O}}{1 \text{ mL H}_2\text{O}} \text{ x} \quad \frac{80 \text{ g NaNO}_3}{100 \text{ g H}_2\text{O}} = 48 \text{ g NaNO}_3$ 

# **Solubility Curves Practice Problems**

**M** Mosston

**Solubility Curves Practice Problems:** 

Whispering the Strategies of Language: An Psychological Quest through Solubility Curves Practice Problems

In a digitally-driven world wherever monitors reign supreme and immediate conversation drowns out the subtleties of language, the profound secrets and mental subtleties hidden within words often go unheard. However, nestled within the pages of **Solubility Curves Practice Problems** a captivating literary treasure pulsing with fresh emotions, lies an extraordinary quest waiting to be undertaken. Published by a skilled wordsmith, that wonderful opus invites viewers on an introspective trip, delicately unraveling the veiled truths and profound affect resonating within the very fabric of every word. Within the emotional depths of the moving evaluation, we will embark upon a sincere exploration of the book is primary themes, dissect its charming writing style, and succumb to the powerful resonance it evokes heavy within the recesses of readers hearts.

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