## Solubility Rules for Common Ionic Compounds in Water Some Soluble Compounds Exceptions Group IA cations (Li+, Na+, K+. Rb+, No Exceptions

Cs+), NH4+

NO<sub>3</sub>. ClO<sub>4</sub>, CH<sub>3</sub>COO, ClO<sub>3</sub>

Cl., Br : 1

Some Insoluble Compounds

SO42-

PO43-

52-

CO32-OH.

Hydroxides of Group 1A cations, NH<sub>4</sub>+,

Ca2+, Sr2+, Ba2+

Sr2+, Ba2+

No Exceptions

Halides of Ag+, Hg22+, Pb2+

Sulfates of Sr2+. Ba2+, Hg22+, Pb2+

Sulfides of Group 1A cations, NH4+, Ca2+,

Carbonates of Group 1A cations and NH<sub>4</sub>+

Phosphates of Group 1A cations and NH<sub>4</sub><sup>+</sup>

# **Solubility Rule And Mystery Solutions Grade 11**

F Rizvi

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