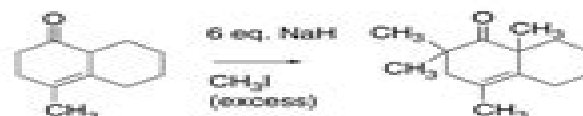


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through γ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH₂ in liquid NH₃. The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH₂ in liquid NH₃.



- e. The structure of the nitrile anion intermediate is shown above. The nitrile anion is stabilized by resonance with the indole ring. The nitrile anion is formed by deprotonation of the nitrile group with NaNH₂ in liquid NH₃. The nitrile anion is then alkylated with PhCH₂Cl to form the final product.
- f. The structure of the nitrile anion intermediate is shown above. The nitrile anion is stabilized by resonance with the indole ring. The nitrile anion is formed by deprotonation of the nitrile group with NaNH₂ in liquid NH₃. The nitrile anion is then alkylated with PhCH₂Cl to form the final product.



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