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CHAPTER 2
ATOMS, MOLECULES, AND IONS

Chapter Learning Goals for Students

- | | |
|--------------------|---|
| Section 2.1 | Use symbols to represent element names. |
| Section 2.2 | Identify the location of metals, nonmetals, and semimetals on the periodic table.

Indicate the atomic number, group number, and period number for an element whose position in the periodic table is given.

Identify groups as main group, transition metal group, or inner transition metal group. |
| Section 2.3 | Specify the location and give examples of elements in the alkali metal, alkaline earth metal, halogens, and noble gas groups.

Use the properties of an element to classify it as metal, nonmetal, or semimetal and give its location in the periodic table. |
| Section 2.4 | Determine the mass of the products in a reaction using the law of mass conservation. |
| Section 2.5 | Demonstrate the law of multiple proportions using mass composition of two compounds of the same elements. |
| Section 2.6 | Describe Thomson's cathode-ray experiment and what it contributed to the current model of atomic structure.

Describe Millikan's oil drop experiment and what it contributed to the current model of atomic structure. |
| Section 2.7 | Describe Rutherford's gold foil experiment and what it contributed to the current |

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A Gutmann



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 - Chapter 2: Essential Elements of Solution Manual 7th Edition Zumdahl
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 - Chapter 4: Solution Manual 7th Edition Zumdahl in Specific Contexts
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