

Answer Key – Practice – Net Ionic Equations (Th 3/18)

Practice #1

Students identify the spectator ions and then write the net ionic equation.



Reactants

Spectator ions:

Na^+ and Cl^-

Net Ionic Equation:



Products



Reactants

Spectator ions:

Na^+ and Cl^-

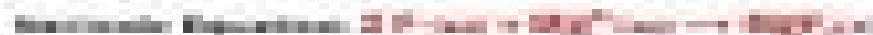
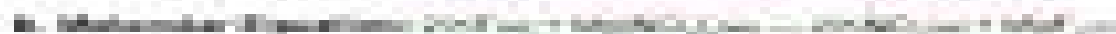
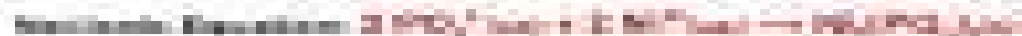
Net Ionic Equation:



Products

Practice #2

Students write complete equations and then write the net ionic equation.



Solubility Rules And Net Ionic Equations Pogil Answer Key

M Carnoy



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