

Understanding Solubility

What is the quantity of solute in a saturated solution at a certain temperature?

Solubility

Suppose you add solute to a given quantity of water in steps through the first 100 ml and repeat this added solute till saturation is reached. You will observe that a certain quantity of solute will dissolve in a certain quantity of water at a certain temperature. This should be a constant for a certain quantity of solute dissolved in a certain quantity of water at a certain temperature. This should be a constant for a certain quantity of solute dissolved in a certain quantity of water at a certain temperature.

Definition: Solubility is the quantity of solute in a saturated solution at a certain temperature.

Example: Solubility of sugar in water at 20°C is 200 g/100 g water.

Example 1: – Finding Solubility

The following table shows the solubility of sugar in water at different temperatures. The solubility of sugar in water at 20°C is 200 g/100 g water. The solubility of sugar in water at 20°C is 200 g/100 g water. The solubility of sugar in water at 20°C is 200 g/100 g water.

Temp.	Experiment 1 in 100 g water		Experiment 2 in 100 g water		Experiment 3 in 100 g water	
	Amount of solute added (g)	Amount of solute dissolved (g)	Amount of solute added (g)	Amount of solute dissolved (g)	Amount of solute added (g)	Amount of solute dissolved (g)
10	100	100	100	100	100	100
20	100	100	100	100	100	100
30	100	100	100	100	100	100
40	100	100	100	100	100	100
50	100	100	100	100	100	100
60	100	100	100	100	100	100
70	100	100	100	100	100	100
80	100	100	100	100	100	100
90	100	100	100	100	100	100
100	100	100	100	100	100	100

1. Solubility of sugar in water at 20°C is 200 g/100 g water. The solubility of sugar in water at 20°C is 200 g/100 g water.

2. Solubility of sugar in water at 20°C is 200 g/100 g water. The solubility of sugar in water at 20°C is 200 g/100 g water.

3. Solubility of sugar in water at 20°C is 200 g/100 g water. The solubility of sugar in water at 20°C is 200 g/100 g water.

Solubility Pogil Answers

Thirumalaisamy P.
Velavancorresponding



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