

LAB 13: SOLUBILITY PRODUCT CONSTANT



Goals

1. Learn of solubility product constants
2. Estimate a solubility constant
3. Observe that a solubility constant is a special case of an equilibrium constant

Materials and Equipment

Reagents:

0.1 M calcium nitrate, $\text{Ca}(\text{NO}_3)_2$

0.1 M sodium oxalate, $\text{Na}_2\text{C}_2\text{O}_4$

Pipet, mini

Toothpick, plastic

Reaction plate, 96 well

Materials Not Included

A sheet of black paper

Distilled water

Introduction

Ionic compounds, such as salts, dissociate to form ions when they dissolve in water. An example is sodium chloride:



Solid NaCl dissociates to become Na^+ and Cl^- ions. The Na^+ cations are attracted to the negative "ends" of water molecules, and the Cl^- anions are attracted to the positive "ends" of water molecules.

In other compounds the attraction between the cations and the anions is stronger, and there is little dissociation. These are sparingly or slightly soluble salts. An example is silver chloride, AgCl . A beaker of this solution often has solid AgCl crystals in the bottom. The crystals are

in equilibrium with the Ag^+ and Cl^- ions that are in the solution:



In this reaction, the rate of AgCl dissolving is equal to the rate of Ag^+ and Cl^- ions combining to form solid AgCl . In other words, the reaction is going both ways at the same time and at the same rate. It is said to be in equilibrium. The equilibrium constant is:

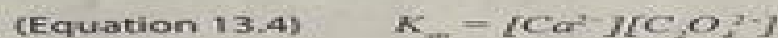


$\text{AgCl}(s)$ is not considered in the equilibrium constant because it is a solid.

We call this special equilibrium constant the solubility product constant, K_{sp} .

Some of the beautiful structures you may have seen in caves are composed of sparingly soluble calcium carbonate. This material slightly dissolves in water, and later precipitates when the water in the solution evaporates or the solution cools.

Another sparingly soluble salt is calcium oxalate, which sometimes precipitates in our kidneys as kidney stones. These can be very painful and often require medical treatment. We will study calcium oxalate, CaC_2O_4 , in this experiment. The solubility product constant for this is:



Solubility Lab Answer Key

Tobias Bleicker

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Solubility Lab Answer Key:

The book delves into Solubility Lab Answer Key. Solubility Lab Answer Key is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Solubility Lab Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Solubility Lab Answer Key
- Chapter 2: Essential Elements of Solubility Lab Answer Key
- Chapter 3: Solubility Lab Answer Key in Everyday Life
- Chapter 4: Solubility Lab Answer Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Solubility Lab Answer Key. The first chapter will explore what Solubility Lab Answer Key is, why Solubility Lab Answer Key is vital, and how to effectively learn about Solubility Lab Answer Key.
3. In chapter 2, this book will delve into the foundational concepts of Solubility Lab Answer Key. The second chapter will elucidate the essential principles that need to be understood to grasp Solubility Lab Answer Key in its entirety.
4. In chapter 3, this book will examine the practical applications of Solubility Lab Answer Key in daily life. The third chapter will showcase real-world examples of how Solubility Lab Answer Key can be effectively utilized in everyday scenarios.
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6. In chapter 5, the author will draw a conclusion about Solubility Lab Answer Key. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Solubility Lab Answer Key.

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