Solubility and Concentration

Match the following. Write the correct letter in the blank provided.

| 1. Solute B | A. Tells you how much selvent is dissolved in the solution- |
|--------------------|---|
| 2. Solvent _E | B. The part that dissolves into the solution |
| 3. Dissolve | C. Increase solubility by moving melecules faster |
| 4. Stirring _c | D. When comething is eeen to disappear inside the solvent |
| 5. Concentration_A | E. The part that dissolves something |

Determine if the following statements are true or false.

a. Solubility increases if you stir a solute into a solution.

b. All solutes have the same solubility in a given solvent. F

c. There is a limit on the amount of solute that can dissolve in a given solvent. _____

d. You can dissolve additional solutes in an unsaturated solution.

3. Give three ways to make sugar dissolve faster. Heat Stirring and pcreasing surface area

4. Which one will dissolve faster? Circle the correct answer.

a. Powdered sugar or granulated sugar? b. In hot water or in cold water?

c. Stirred or not stirred? d. Large particles or small particles?

c. Stirred or not stirred? d. Large particles or small pa

65 g of sugar is dissolved in 750 ml of water. What is the concentration of the solution?
 0.08 g/ml

 Which is more concentrated: 34 g of salt dissolved in 100 ml of water or 100 g of salt in 1500 ml of water?
 34 g of Salt in 100ml of water

7. If the solubility of salt in water was determined to be 0.5 g/ml, would a solution with 50 g of salt in 150 ml of water be considered saturated?

Solubility And Concentration Answer Key

John Olmsted, Greg Williams, Robert C. Burk

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