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questions what affects how a gas behaves what are the gas laws what affects the behavior of a gas gases behave differently than solids or liquids gas particles have a large amount of space between them

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web the amount of space that matter fills measured in cubic centimeters cm3 milliliters ml liters l gas particles move and fill the space available this measure of gas is the same as the measure of its container a measure of the average energy of random motion of the particles of a substance

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web jun 9 2016 chapter 3 section 3 behavior of gases show full text are fluids have no definite shape or volume and they expand to fill their container particles move rapidly in all directions molecules are in constant motion and collide with each other and walls of the container low density because particles are far apart are compressible