

CHEM 120 Exam 2: Chapters 5,6,7,8 (With Answer Key) - Chamberlain College of Nursing

Examination Study Guide 2023

Exam 2

Total Points: 60

PART 1: Multiple Choice questions: Q 1-24. (2 point each)

- Chemical reactions (equations) show the following
 - Reactants
 - Products
 - Physical state of each (gas or liquid or solid)
 - All of the above
- The following is true for a mole
 - It is also known as Avogadro's number
 - It contains 6.022×10^{23} molecules
 - It contains mass equivalent to molecular weight
 - All of the above are true
- In balancing chemical equations, the coefficients are used to
 - Show number of moles
 - Show volume of each in liters
 - Show amount of each in grams
 - Show temperature at which the substance is stable
- $2\text{H}_2 + \text{O}_2 \longrightarrow 2\text{H}_2\text{O}$ means that
 - 2 grams of H_2 g and 1 gram of O_2 gas combine to form 2 grams of water.
 - 2 moles of H_2 g and 1 mole of O_2 gas combine to form 2 moles of water.
 - 2 grams of H_2 g and 1 gram of O_2 gas combine to form 2 moles of water.
 - 2 molecules of H_2 g and 1 molecule of O_2 gas combine to form 2 molecules of water.
- $2\text{H}_2 + \text{O}_2 \longrightarrow 2\text{H}_2\text{O}$. In this equation, correct method to calculate molecular mass of water is
 - $2[2(\text{mass of hydrogen}) + \text{mass of oxygen}]$
 - $2(\text{mass of hydrogen gas}) + \text{mass of oxygen gas}$
 - $2(\text{mass of hydrogen}) + \text{mass of oxygen}$

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OECD



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OECD Series on Testing and Assessment Guidance Document on Aquatic Toxicity Testing of Difficult Substances and Mixtures OECD,2019-05-16 This document provides the second edition of Guidance Document 23 on Aqueous Phase Aquatic Toxicity Testing of Difficult Test Chemicals **Resources in education** ,1986-07 Environmental Toxicology Mr. Rohit Manglik,2023-07-23 Toxic substances and environmental impact Includes toxicity exposure and health effects essential for environmental health and risk assessment studies **Living by Chemistry (2018 Update)** Angelica M. Stacy,2019-03-07 Designed to help all students to learn chemistry Living by Chemistry is a full year high school curriculum that incorporates science practices with a guided inquiry approach Students of all levels will gain a deep understanding of chemistry with this program With Living by Chemistry students learn chemistry in the same way that chemists work by asking questions collecting evidence and thinking like scientists Living by Chemistry is the product of a decade of research and development in high school classrooms focusing on optimizing student understanding of chemical principles Author Angelica Stacy assisted in the development of the NGSS standards and served on the AP Chemistry redesign committee She designed Living by Chemistry as an introduction for students who will take AP Chemistry or additional college classes The curriculum was developed with the belief that science is best learned through first hand experience and discussion with peers Guided inquiry allows students to actively participate in and become adept at scientific processes and communication These skills are vital to a students further success in science as well as beneficial to other pursuits Formal definitions and formulas are frequently introduced after students have explored scrutinized and developed a concept providing more effective instruction LBCs innovative curriculum offers much more than traditional programs To help engage students of all levels the curriculum provides a variety of learning experiences through activities discussions games demos lectures labs and individual work **Environmental Toxicology and Chemistry** ,2006 Resources for Teaching Middle School Science Smithsonian Institution,National Academy of Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also

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