Table 18.2

Salt	K _{sp}	Salt	K _{sp}	Salt	K _{sp}
Halides	0,50	Sulfates	250	Hydroxides	
AgCI	1.8×10^{-10}	PbSO ₄	6.3 × 10 ⁻⁷	Al(OH) ₃	3.0×10^{-34}
AgBr	5.0×10^{-13}	BaSO,	1.1×10^{-19}	Zn(OH) ₂	3.0×10^{-16}
Agl	8.3×10^{-17}	CaSO ₄	2.4×10^{-6}	Ca(OH) ₂	6.5×10^{-4}
PbCl ₂	1.7 × 10 ⁻⁵	Sulfides	1 Distriction of the Control of the	Mg(OH) ₂	7.1×10^{-12}
PbBr ₂	2.1×10^{-6}	NiS	4.0×10^{-29}	Fe(OH) ₂	7.9×10^{-16}
Pbl ₂	7.9×10^{-9}	CuS	8.0×10^{-37}	Carbonates	
PbF ₃	3.6×10^{-6}	Ag ₂ S	8.0×10^{-61}	CaCO ₃	4.5×10^{-9}
CaF ₂	3.9×10^{-11}	ZnS	3.0×10^{-23}	SrCO ₃	9.3×10^{-10}
Chromates	no de deconoción de	FeS	8.0×10^{-19}	ZnCO ₃	1.0×10^{-10}
PbCrO,	1.8×10^{-14}	CdS	1.0×10^{-37}	Ag ₂ CO ₃	8.1×10^{-12}
Ag ₂ CrO ₄	1.2×10^{-12}	PbS	3.0×10^{-38}	BaCO,	5.0×10^{-9}



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Thomas Griffiths

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