- Define solubility.
- 2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve?
- II. Saturation points are given per how much of the solvent?
- III. What is the saturation point of NaNO3 at 10°C, 40°C, & 80°C?
- IV. How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
- 10°C = \_\_\_\_\_ 30°C = \_\_\_\_\_ 90°C = \_\_\_\_\_
- V. Which salt is the least soluble in water?
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C?
- VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water?\_\_\_
- VIII. Which salt shows the least change in solubility from 0° to 100° C? \_\_\_\_\_
- IX. Which substance shows a decrease in solubility from 0° to 100° C?
- X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? \_\_\_\_\_\_

# **Solubility Temperature Graphs Chapter 1worksheet Answers**

**L** Towne

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