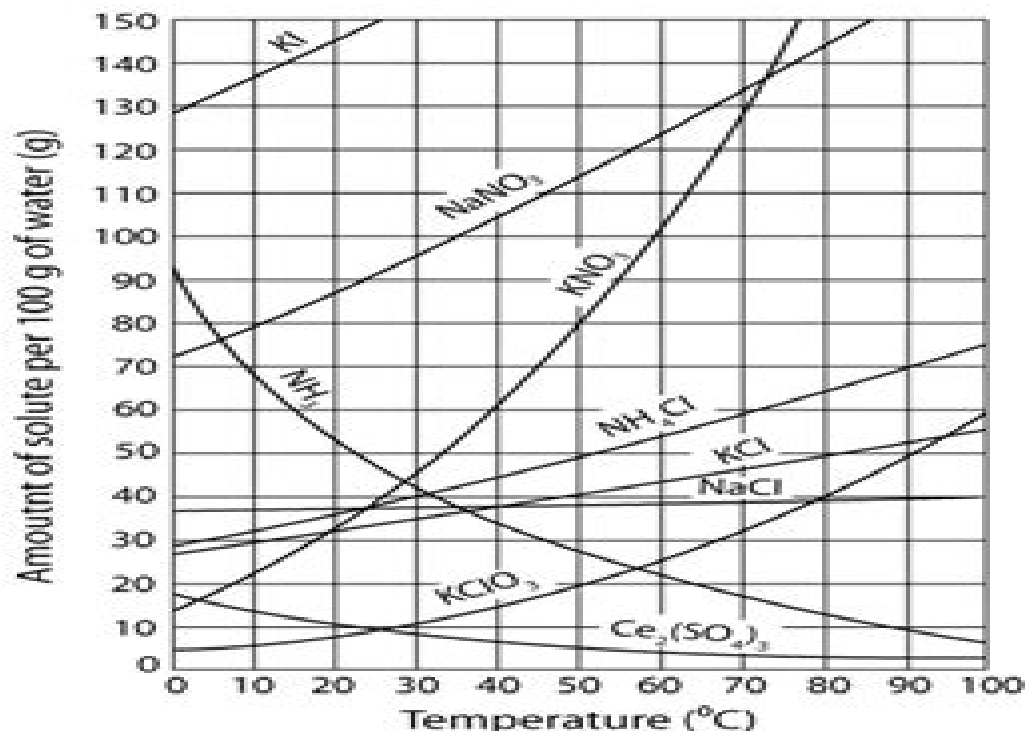


Practice Using Solubility Curve

The image below represents the solubility curves of several compounds. Answer the given questions.



1. List the substances whose solubility decreases as temperature increases. _____
2. Which substance is least affected by temperature changes? _____
3. How many grams of ammonium chloride (NH₄Cl) forms a saturated solution at 50°C? _____
4. Which two substances have the same solubility at approximately 78°C? _____
5. Which compound is least soluble in water at 10°C? _____
6. What is the solubility of KCl at 25 °C? _____
7. What is the solubility of Ce₂(SO₄)₃ at 50 °C? _____
8. How many grams of KNO₃ can be dissolved at 50°C? _____

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century This was more a technology than a science largely because trial and error was the only method to obtain good results However the situation is now changing There are so many appealing hidden findings in the process of crystallization Historically crystallization began in an old laboratory in academia and then gradually shifted to industry Now it is making its comeback in academia due to several new research branches trying to discover what is going on during the crystallization process I believe this old of science is now growing as a result of the wonderful coupling between industry and academia I read a prepublication draft of Novel Optical Resolution Technologies and found that each one of these general characteristics of science had a reality and sharpness that I had not expected While it was a sheer delight to revisit each of these triumphs guided by the wise insights and analyses found throughout the book There is a good balance between the underlying historical material and the design and execution aspects of each topic Chemistry Calculations for Beginners John Obimakinde, Samuel

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Pharmaceutical Sciences Patrick J. Sinko, 2023-02-08 Consistently revised and updated for more than 60 years to reflect the most current research and practice Martin's Physical Pharmacy and Pharmaceutical Sciences 8th Edition is the original and most comprehensive text available on the physical chemical and biological principles that underlie pharmacology and the pharmaceutical sciences An ideal resource for PharmD and pharmacy students worldwide teachers researchers or industrial pharmaceutical scientists this 8th Edition has been thoroughly revised enhanced and reorganized to provide readers with a clear consistent learning experience that puts essential principles and concepts in a practical approachable context Updated content reflects the latest developments and perspectives across the full spectrum of physical pharmacy and a new full color design makes it easier than ever to discover distinguish and understand information providing users the most robust support available for applying the elements of biology physics and chemistry in work or study **Crystallization and Materials**

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