

## SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degrees Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases.  $\text{NaCl}$  and  $\text{CaSO}_4$
5. Which substance is least affected by temperature changes?  $\text{NaCl}$
6. How many grams of ammonium chloride ( $\text{NH}_4\text{Cl}$ ) at  $50^\circ\text{C}$ ? 50g
7.  $\text{NaCl}$  and  $\text{KCl}$  have the same solubility at approximately  $75^\circ\text{C}$ .
8. Which compound is least soluble in water at  $10^\circ\text{C}$ ?  $\text{KClO}_3$
9. How many grams of  $\text{KNO}_3$  can be dissolved at  $50^\circ\text{C}$ ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
  - a. 40g of  $\text{NaNO}_3$  in 100 g of water at  $30^\circ\text{C}$ . saturated
  - b. 60g of  $\text{KClO}_3$  in 100 g of water at  $90^\circ\text{C}$ . supersaturated
11. How many grams of sodium chloride,  $\text{NaCl}$  are required to saturate 100 grams of water at  $100^\circ\text{C}$ ? 40g
12. How many grams of  $\text{NaNO}_3$  are required to saturate 100 grams of water at  $75^\circ\text{C}$ ? 140g
13. How many grams of  $\text{KI}$  will saturate water at  $20^\circ\text{C}$ ? 30g
14. At what temperature would 25g of potassium chlorate ( $\text{KClO}_3$ ) dissolve?  $85^\circ\text{C}$
15. At what temperature would 60g of  $\text{NH}_4\text{Cl}$  dissolve?  $35^\circ\text{C}$
16. 80-g  $\text{NaNO}_3$  is prepared at  $30^\circ\text{C}$ .

# Solubility Answers Key

**M Mark**



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## **Table of Contents Solubility Answers Key**

1. Understanding the eBook Solubility Answers Key
  - The Rise of Digital Reading Solubility Answers Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Solubility Answers Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Solubility Answers Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Solubility Answers Key
  - Personalized Recommendations
  - Solubility Answers Key User Reviews and Ratings
  - Solubility Answers Key and Bestseller Lists

5. Accessing Solubility Answers Key Free and Paid eBooks
  - Solubility Answers Key Public Domain eBooks
  - Solubility Answers Key eBook Subscription Services
  - Solubility Answers Key Budget-Friendly Options
6. Navigating Solubility Answers Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Solubility Answers Key Compatibility with Devices
  - Solubility Answers Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Solubility Answers Key
  - Highlighting and Note-Taking Solubility Answers Key
  - Interactive Elements Solubility Answers Key
8. Staying Engaged with Solubility Answers Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Solubility Answers Key
9. Balancing eBooks and Physical Books Solubility Answers Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Solubility Answers Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Solubility Answers Key
  - Setting Reading Goals Solubility Answers Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Solubility Answers Key
  - Fact-Checking eBook Content of Solubility Answers Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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