

USING CORRECT UNITS

Name _____

For each of the following commonly used measurements, indicate its symbol. Use the symbols to complete the following.

_____ milliliter	_____ milligram	_____ liter	_____ centimeter
_____ kilogram	_____ millimeter	_____ kilometer	_____ gram
_____ meter	_____ millisecond	_____ microgram	_____ nanometer

1. Colas may be purchased in two or three _____ bottles.
2. The mass of bowling ball is 7.25 _____.
3. The length of the common housefly is about 1 _____.
4. The mass of a paper clip is about 1 _____.
5. One teaspoon of cough syrup has a volume of 5 _____.
6. The speed limit on the highway is usually 106.6 _____ /h or 29 _____ /s.
7. The length of the small intestine in man is about 6.25 _____.
8. Viruses such as AIDS, polio and flu range in length from 17 to 1000 _____.
9. Adults require 1,000 _____ of calcium to meet the U.S. RDA.
10. In a vacuum, light can travel 300 km in 1 _____.
11. The mass of a proton is 1.67×10^{-18} _____.
12. Blue light has a wavelength of about 500 _____.
13. One mole of oxygen gas at STP occupies 22.4 _____.
14. Myoglobin, a protein that stores oxygen, has a mass of 2.98×10^{-16} _____.
15. Buttery popcorn contained in a large 1 _____ bowl has a mass of about 50 _____ of fat and about 650 calories.
16. The dying comet fragments that continued to batter Jupiter travel at speeds of about 58,117 _____ / _____ or 130,000 miles per hour.
17. The human heart has a mass of about 1.05 _____.
18. Stand with your arms raised out to your side. The distance from your nose to your outstretched middle finger is about 1 _____.
19. The body mass of a flea is about 0.5 _____ and it can jump about 20 _____ high.
20. On a statistical basis, smoking a single cigarette lowers your life expectancy by 642,000 _____ or 10.7 minutes.

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Spectrum



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Chemistry and Physics for Nurse Anesthesia, Second Edition David Shubert, David C. Shubert, John Leyba, 2013-03-15. Praise for the first edition: A welcome addition to the reference materials necessary for the study of nurse anesthesia. The textbook is divided into logical, easy-to-use sections that

cover all areas necessary for the practice of nurse anesthesia This is a text that is easy to read and able to be incorporated into any nurse anesthesia chemistry and physics course I would recommend this textbook to any program director Anthony Chipas PhD CRNA Division Director Anesthesia for Nurses Program Medical University of South Carolina Nurse anesthesia students will welcome the second edition of this text designed for the combined course in chemistry and physics that is required for this program It is written in a clear conversational style to counteract the trepidation that often accompanies the study of chemistry and physics and includes only those core scientific concepts that relate to clinical anesthesia application Numerous illustrations demonstrate how the scientific concepts relate directly to their clinical application in anesthesia and plentiful case studies exemplify and reinforce basic concepts Review question at the end of each chapter facilitate self assessment This second edition offers numerous features that will further assist students with understanding and mastery of the material These new features are the direct result of knowledge gained from on line and traditional classroom teaching experiences They include chapter summaries additional questions and answers at the end of each chapter specific to nurse anesthesia end of chapter summaries and lists of formulas and constants discussed in the book Fifteen videos vividly demonstrate the key principles of the chemistry and physics of nurse anesthesia Corresponding to various sections of the book they supplement and illustrate text content Also available are revised PowerPoint slides for faculty use The first edition of this popular text is currently being used by eight nurse anesthesia programs throughout the United States and many additional programs plan to adopt the second edition New to the Second Edition Emphasizes content in chemistry and physics that relates specifically to anesthesia with a strong focus on gases Includes case studies to illustrate and reinforce knowledge Provides additional end of chapter problems focused on anesthesia Relates core scientific concepts to clinical anesthesia application Offers fifteen videos demonstrating key principles of the physics and chemistry of nurse anesthesia

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