

CHAPTER

02

SOLUTION

1. **Solution** : A homogeneous mixture of two or more substances is known as solution.
2. **Solute** : The substance present in smaller amount in a solution is called solute.
3. **Solvent** : The substance present in larger amount in a solution is called solvent.
4. **Binary solution** : Those solutions which contain two components are called binary solutions, e.g., salt solution, benzene and toluene.
5. **Ternary solutions** : Those solutions which contain three components are called ternary solutions, e.g., ethanol + water + acetic acid.
6. **Aqueous solution** : When solute is dissolved in water, it is called aqueous solution, e.g., sugar solution, ethanol in water.
7. **Non-aqueous solution** : When solute is dissolved in solvent other than water, it is called non-aqueous solution, e.g., iodine dissolved in alcohol (Tincture of iodine).
8. **Weight Fraction** = $\frac{\text{weight of solute}}{\text{weight of solution}}$
9. **Weight percentage**

$$= \frac{\text{weight of solute}}{\text{weight of solution}} \times 100$$

Weight of solute present in 100 grams of solution is called weight percentage

Volume of solute present in 100 ml of the solution is called volume percentage
10. **Volume percentage**

$$= \frac{\text{Volume of solute}}{\text{Volume of solution}} \times 100$$
11. **Volume Fraction** = $\frac{\text{Volume of solute}}{\text{Volume of solution}}$
12. **Molarity (M)**

$$= \frac{\text{No of moles of solute}}{\text{Volume of solution in liters}}$$

$$(\text{or}) M = \frac{W}{GMW} \times \frac{1}{V(\text{lit})}$$
13. **Molarity** = $\frac{\text{No of moles of solute}}{\text{Volume of solution in ml}} \times 1000$
14. **Molarity** = $\frac{\text{Weight of solute}}{\text{GMWt of solute}} \times \frac{1000}{V \text{ in ml}}$

$$(\text{or}) M = \frac{W}{GMW} \times \frac{1000}{V(\text{ml})}$$
15. **Weight of solute** = $\frac{M \times GMW \times V \text{ in ml}}{1000}$
16. **Molarity for dilute solutions** : $M_1 V_1 = M_2 V_2$
17. **For neutralisation reactions** : $\frac{M_1 V_1}{n_1} = \frac{M_2 V_2}{n_2}$
18. **For mixed solution molarity**

$$= \frac{M_1 V_1 + M_2 V_2 + \dots}{V_1 + V_2 + \dots}$$

Solutions Chapter

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Massive MIMO in Practice Henrik Asplund, George Jöngren, Jonas Karlsson, Fredric Kronestedt, Erik Larsson, David Astely, Peter von Butovitsch, Thomas Chapman, Sebastian Faxér, Mattias Frenne, Farshid Ghasemzadeh, Måns Hagström, Billy Hogan, 2025-05-30 Massive MIMO in Practice From 5G 5G Advanced to 6G 2nd edition of the previously titled Advanced Antenna Systems for 5G Network Deployments Bridging the Gap between Theory and Practice provides a theoretical introduction to Massive MIMO as well as presenting how it can meet network performance requirements for commercial deployment Features include A thorough understanding of Array antennas and how they can be used for beamforming null forming and support for MIMO features Massive MIMO features and how they work with a particular focus on mobile networks and the specifics relevant for mobile network operation e.g. characteristics of the radio channel in different environment and how Massive MIMO solutions adapt to these A detailed walk through of the 3GPP physical layer support for Massive MIMO solutions the background for this support and how it can be used in mobile networks Explanations of what

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Enterprise Master Data Management Allen Dreibelbis,Eberhard Hechler,Ivan Milman,Martin Oberhofer,Paul van Run,Dan Wolfson,2008-06-05 The Only Complete Technical Primer for MDM Planners Architects and Implementers Companies moving toward flexible SOA architectures often face difficult information management and integration challenges The master data they rely on is often stored and managed in ways that are redundant inconsistent inaccessible non standardized and poorly governed Using Master Data Management

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prompts that help you reflect on an issue and in some cases draw on your own classroom experience prior to reading more about it Linking To Your Own Instruction sections help you implement the five practices with confidence in your own instruction The book and companion website provide an array of resources including planning templates sample lesson plans completed monitoring tools and mathematical tasks Enhance your fluency in the five practices to bring powerful discussions of mathematical concepts to life in your classroom

Polyparallelogram Puzzles and Tiling Problems Raymond R. Fletcher III, A mathematical puzzle book which involves labeling vertices of a plane graph with elements from an abelian group G such that the labels applied to the bounding vertices of each region have the same sum in G The plane graphs are those formed by the edges and vertices of various types of parallelograms fitted together edge to edge We call these polyparallelograms The polyparallelogram puzzles are organized into the following categories according to the types of parallelograms involved i polysquares or polyominoes ii 60 120 rhombs iii 36 144 and 72 108 rhombs Penrose rhombs iv various lattice parallelograms The reader is introduced to the theory of polyparallelogram tilings of polygons and challenged with problems and research questions at the end of each Chapter All that is required of the elementary theory of graphs and abelian groups is included in the text Also included are a variety of pure tiling problems involving nonparallelogram lattice quadrilaterals Polyparallelogram Puzzles and Tiling Problems is a sequel to Polycubes Triangulations and Polyhexes over $\mathbb{Z}_n$ and features a similar type of puzzle but is more engaging mathematically

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