## lons

When ions are formed electrons are gained (anion) or lost (cation) to form a stable arrangement.

E.g.

P<sup>3-</sup>, S<sup>2-</sup>, Cl<sup>-</sup> & Ar are all 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>2</sup> 3s<sup>2</sup> 3p<sup>6</sup>

They are Isoelectronic

# **Small Scale Invest Electron Configurations Atoms And Ions**

**J Spring** 

#### **Small Scale Invest Electron Configurations Atoms And Ions:**

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