



Titration

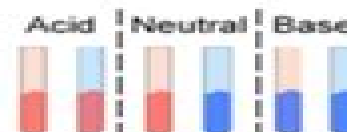
Answer Key

Vocabulary: acid, analyte, base, dissociate, equivalence point, indicator, litmus paper, molarity, neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

There are several definitions of acids and bases. According to the Brønsted-Lowry definition, an **acid** is a substance that is capable of donating a proton to another substance. A **base** is a substance that accepts protons. When an acid and a base are combined, the acid is **neutralized** as the base accepts the protons produced by the acid.



One way to determine if a solution is acidic or basic is to use **litmus paper**, as shown above. There are two types of litmus papers: red and blue.

How does litmus paper indicate an acid? *Both strips turn red.*

How does litmus paper indicate a neutral substance? *The red strip stays red and the blue strip stays blue.*

How does litmus paper indicate a base? *Both strips turn blue.*

Gizmo Warm-up

Litmus is an example of an **indicator**, a substance that changes color depending on its **pH** (pH is a measure of the concentration of protons, or H^+ ions). In the *Titration* Gizmo, you will use indicators to show how acids are neutralized by bases, and vice versa.

To begin, check that **1.00 M NaOH** is selected for the **Burette**, **Mystery HBr** is selected for the **Flask**, and **Bromthymol blue** is selected for the **Indicator**.



1. Look at the flask. What is the color of the bromthymol blue indicator? *Yellow*

2. What does this tell you about the pH of the solution in the flask? *pH is below 6.0*

Solutions with a pH below 7.0 are acidic, while those with a pH above 7.0 are basic.

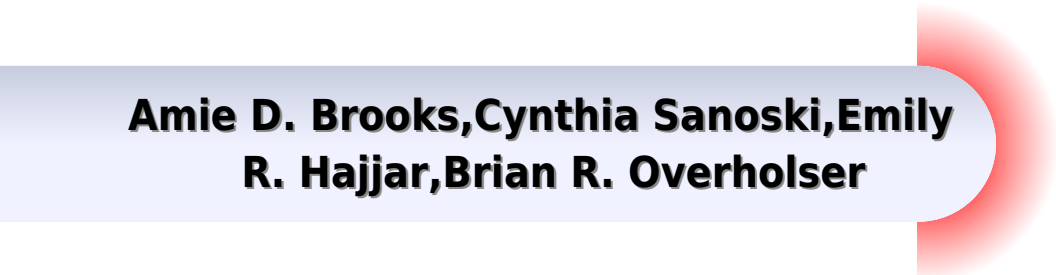
3. Move the slider on the burette to the top to add about 25 mL of NaOH to the flask. What happens, and what does this tell you about the pH of the flask?

The solution turns blue, indicating the pH is now above 7.6.



Titration Virtual Activity Answer Key

**Amie D. Brooks, Cynthia Sanoski, Emily
R. Hajjar, Brian R. Overholser**



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Environmental Health Perspectives ,2003-07

Freshney's Culture of Animal Cells Amanda Capes-Davis,R. Ian Freshney,2021-02-17 FRESHNEY S CULTURE OF ANIMAL CELLS THE NEW EDITION OF THE LEADING TEXT ON THE BASIC METHODOLOGY OF CELL CULTURE FULLY UPDATED TO REFLECT NEW APPLICATIONS INCLUDING IPSCS CRISPR AND ORGAN ON CHIP TECHNOLOGIES Freshney s Culture of Animal Cells is the most comprehensive and up to date resource on the principles techniques equipment and applications in the field of cell and tissue culture Explaining both how to do tissue culture and why a technique is done in a particular way this classic text covers the biology of cultured cells how to select media and substrates regulatory requirements laboratory protocols aseptic technique experimental manipulation of animal cells and much more The eighth edition contains extensively revised material that reflects the latest techniques and emerging applications in cell culture such as the use of CRISPR Cas9 for gene editing and the adoption of chemically defined conditions for stem cell culture A brand new chapter examines the origin and evolution of cell lines joined by a dedicated chapter on irreproducible research its causes and the importance of reproducibility and good cell culture practice Throughout the book updated

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