

MAIL PIECE DESIGN AND FOLDING REQUIREMENTS

Effective 1/26/14. Compliments of AlphaGraphics

DESIGN REQUIREMENTS

LETTERS AND POSTCARDS

Bar code must be within 4" of mail piece bottom, .5" from right or left edge and .25" away from printed graphics. Entire address block and bar code must be above .625" from bottom of piece. Please allow at least 1.5"H x 3.5"W for mail info (.25" from printed graphics and .5" from edge of sheet is included in this measurement).

FLATS

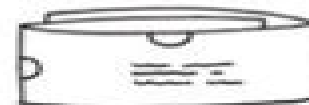
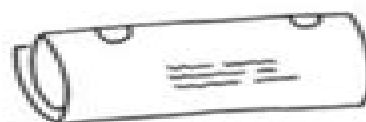
Bar code must be within 4" of mail piece top, .5" from right or left edge and .25" away from printed graphics. Entire address block and bar code cannot be inline with the return address or indicia and must be on the right or left side, not in the center. Please allow at least 1.5"H x 3.5"W for mail info (.25" from printed graphics and .5" from edge of sheet is included in this measurement).

ASPECT RATIO LETTERS AND POSTCARDS ONLY

To qualify for automated bulk mail rates, the aspect ratio (length divided by height) must be greater than 1.3 and less than 2.5.

FOLDING

If the piece is a tri-fold or folded self mailer the final fold must be on the bottom and the second fold must always be at the right leading edge. For tri-folded mailers, the mailing address must be on the middle panel, with the final folded panel creating the non-address side of the mail piece, as shown on the right. If the fold(s) are on the side, the final fold must be on the right leading edge.



REV_2.08.04

Usps Mailpiece Design Consultant Study Guide

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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web mar 27 2023 a the lewis structure for water is represented by the given option a in this structure the oxygen atom is in the center and is bonded to two hydrogen atoms the oxygen atom has two lone pairs of electrons this structure follows the octet rule where each atom has a full valence shell with eight electrons

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web introduction the physical and chemical behavior of compounds depends on the types of bonds that exist within the molecule and on the overall shape of the molecules the branch of chemistry that investigates how molecules are put together is

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web lewis structures represent the valence electrons present in the molecule as lone pairs unshared electrons drawn as dots or bonds shared electrons drawn as lines procedure for drawing lewis structures add up the total number of valence electrons that each atom contributes to the molecule polyatomic ion

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web the pcl 3 lewis structure has the typical case of phosphorus p in the center with 3 bonds to 3 other atoms phosphorus is from the same column as nitrogen in the periodic table meaning that p and n generally have the same bonding structure note the lone pair dots without bonds on top of p just like for n in the previous example for nh 3 chlorine cl

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web step 1 figure out how many electrons the molecule must have based on the number of valence electrons in each atom when drawing the structure of an ion be sure to add subtract electrons to account for the charge step 2 connect the atoms to each other with single bonds to form a skeleton structure

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web lewis structures a lewis structure is a representation of covalent molecules or polyatomic ions where all the valence electrons are shown distributed about the bonded atoms as either shared electron pairs bond pairs or unshared electron pairs lone pairs a shared pair of electrons is represented as a short line a single bond

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web this exercise required us to take the formulas in the data table and calculate the number of valence electrons using the periodic table draw its lewis structure on paper build the molecular model and record a photo list the number of bonds and lone pairs surrounding the central atom and lastly identify the structure geometry

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