Balancing Equations 12	
lame	Period
. BClg + P4 + H2 → BP + HCl	
. $C_2H_2Cl_4 + Co(OH)_2 \rightarrow C_2HCl_3 + CoCl_2 + H_2Cl_3$	D
3. (NH4)2Cr2O7 → N2 + Cr2O3 + H2O	
4. Zn3Sb2 + H2O → Zn(OH)2 + SbH3	
5. HCIO4 + P4O10 -> H3PO4 + CI2O7	
6. C6H5C1 + SICI4 + Na → (C6H5)4S1 + NaC1	
7. Sb <sub>2</sub> S <sub>3</sub> + HCl → H <sub>3</sub> SbCl <sub>6</sub> + H <sub>2</sub> S	
8. IBr + NH3 -> NH3 + NH4Br	
9. KrF2 + H2O -> Kr + O2 + HF	-
10. Na2CO3 + C + N2 → NaCN + CO	
11. Fe (CO) 5 + 1) = OH -> Nog Fe (CO) 4 +	Uaz (03 + H2U
E.C. H. PO. + (UH4), UOO4 + HNO3 -> (NH4), PO4 - 12	Ma 03 + NH4 NO3

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Robert Dąbrowski

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