## Word Equations

Write the skeleton equations for each of the following chemical reactions and then balance them:

 When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.

$$BeCl_{2(aq)} + 2 AgNO_{3(aq)} \rightarrow Be(NO_3)_{2(aq)} + 2 AgCl_{(s)}$$

 When isopropanol (C<sub>3</sub>H<sub>0</sub>O) burns in oxygen, carbon dioxide, water, and heat are produced.

$$2 C_3H_8O_{(1)} + 9 O_{2(g)} \rightarrow 6 CO_{2(g)} + 8 H_2O_{(g)}$$

When dissolved sodium hydroxide reacts with sulfuric acid, aqueous sodium sulfate, water, and heat are formed.

4) When fluorine gas is put into contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.

$$F_{2(g)} + Ca_{(s)} \rightarrow CaF_{2(s)}$$

When sodium metal reacts with iron (II) chloride, iron metal and sodium chloride are formed.

# **Worksheet Word Equations Chemistry A Study Of Matter**

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Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack **Popular Mechanics**, 2000-01 Popular Mechanics inspires instructs and influences readers to help them Price, 1998 master the modern world Whether it's practical DIY home improvement tips gadgets and digital technology information on the newest cars or the latest breakthroughs in science PM is the ultimate guide to our high tech lifestyle and Materials for the Mentally Retarded, Specific Learning Disabled, Emotionally Disturbed National Information Center for Special Education Materials, 1978 Science Spectrum Holt Rinehart & Winston, Holt, Rinehart and Winston Staff,2003-03 The Multimedia and CD-ROM Directory ,1998 **Chemistry, Problem-Solving Worktext** James E. Brady, Frederick Senese, 2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes both chemical and physical that they undergo Fireworks One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light Carbon The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C 2H 2 and O 2 to give CO 2 and H 2O Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms Stormy weather The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet Chemistry, Student Solutions Manual James E. Brady, Joel W. Russell, John R. Holum, 2000-01-26 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical substances and the changes both chemical and physical that they undergo Fireworks One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light Carbon The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs Diamond is not the most stable form of carbon and is created

from other forms of carbon at high temperatures and pressures deep within the earth Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C 2H 2 and O 2 to give CO 2 and H 2O Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms Stormy weather The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet Coordinating Procedural and Conceptual Knowledge to Make Sense of Word Equations: Understanding the Complexity of a 'simple' Chemical Task at the Learner's Resolution Keith Stephen Taber, Pat Bricheno, 2009 Abstract This paper discusses the conceptual demands of an apparently straightforward task set to secondary level students completing chemical word equations with a single omitted term Chemical equations are of considerable importance in chemistry and school students are expected to learn to be able to write and interpret them However it is recognized that many students find them challenging The present paper explores students accounts of their attempts to identify the missing terms to illuminate why working with chemical word equations is so challenging from the learner's perspective 300 secondary age students responded to a 5 item exercise based on chemicals and types of reactions commonly met at school level For each item they were asked to identify the missing term in a word equation and explain their answers This provided a database containing more than a thousand student accounts of their rationales Analysis of the data led to the identification of seven main classes of Chemistry, Student Solutions Manual James E. Brady, John R. Holum, 1996 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving **Chemistry Equations And Answers (Speedy Study Guides)** Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance) Chris and solutions McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these chemical reactions Word Equations and Related Topics INTERNATIONAL WORKSHOP ON WORD EQUATIONS AND RELATED TOPICS.1993 Chemistry: Matter & Change, Study Guide For Content Mastery, Student Edition McGraw Hill, 2001-03-30 Study Guide and Reinforcement Worksheets allow for differentiated instruction through a wide range of

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