Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



- What type of charge does a proton have? Positive (+1)
- What type of charge does a neutron have? Neutral (0)
- What type of charge does an electron have? Negative (-1)
- Which two subatomic particles are located in the nucleus of an atom?
 Protons and neutrons
- 8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
 35
- 9. What is the atomic number of the atom in the diagram above?
- What is the atomic mass/mass number of the atom in the diagram above?
 (11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
- 11. How many protons are in the nucleus of an atom with an atomic number of 15?
 15 *number of protons is the same as the atomic number
- How many electrons are in the nucleus of an atom with an atomic number of 20?
- How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
 Neutrons, atomic mass: 54.938 ~ 55 atomic number: 25....55-25=30
- 14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
 Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
- How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
 Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Worksheet Chapter Atomic Structure

Gurukul

Worksheet Chapter Atomic Structure:

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Table of Contents Worksheet Chapter Atomic Structure

- 1. Understanding the eBook Worksheet Chapter Atomic Structure
 - The Rise of Digital Reading Worksheet Chapter Atomic Structure
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Worksheet Chapter Atomic Structure
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Worksheet Chapter Atomic Structure
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Worksheet Chapter Atomic Structure
 - Personalized Recommendations
 - Worksheet Chapter Atomic Structure User Reviews and Ratings
 - Worksheet Chapter Atomic Structure and Bestseller Lists

- 5. Accessing Worksheet Chapter Atomic Structure Free and Paid eBooks
 - Worksheet Chapter Atomic Structure Public Domain eBooks
 - Worksheet Chapter Atomic Structure eBook Subscription Services
 - Worksheet Chapter Atomic Structure Budget-Friendly Options
- 6. Navigating Worksheet Chapter Atomic Structure eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Worksheet Chapter Atomic Structure Compatibility with Devices
 - Worksheet Chapter Atomic Structure Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Worksheet Chapter Atomic Structure
 - Highlighting and Note-Taking Worksheet Chapter Atomic Structure
 - Interactive Elements Worksheet Chapter Atomic Structure
- 8. Staying Engaged with Worksheet Chapter Atomic Structure
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Worksheet Chapter Atomic Structure
- 9. Balancing eBooks and Physical Books Worksheet Chapter Atomic Structure
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Worksheet Chapter Atomic Structure
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Worksheet Chapter Atomic Structure
 - Setting Reading Goals Worksheet Chapter Atomic Structure
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Worksheet Chapter Atomic Structure
 - Fact-Checking eBook Content of Worksheet Chapter Atomic Structure
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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