# **Atoms and Molecules Worksheet**

Using the Periodic Table located inside your text book, answer the following questions.

Analyzing Data: The information below is from part of the Periodic Table of Elements. Use this information to answer the questions that follow.

IA 1 H		**************************************	2007-0-20-0				VIII 2 He
1	IIA	IIIB 5	IVB 6	VB	VIB	VIIB	4
Li 7	Be 9	B 11	C 12	N 14	8 O 16	F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 CI 35	18 Ar 40

Carbon (C)

1) How many protons, neutrons, and electrons are in each of the following atoms?

Lithium (Li) p = \_\_\_\_\_ n = \_\_\_\_

Magnesium (Mg) p = \_\_\_\_

P-\_\_\_\_

Oxygen (O) p = \_\_\_\_\_ n =

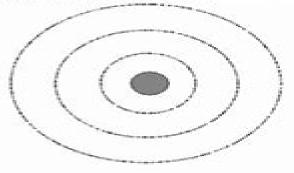
Chlorine (CI)

p = \_\_\_\_\_

n =

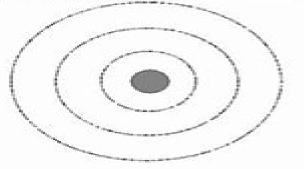
2) Fill in the number of electrons in the correct energy levels below.

Number of Fluorine (F) electrons:



Is Fluorine Stable? \_\_\_\_\_
Explain your answer: \_\_\_\_\_

Number of Neon (Ne) electrons:



Is Neon Stable? \_\_\_\_\_
Explain your answer: \_\_\_\_\_

# **Worksheet Rearranging Atoms**

John Loughran, Amanda Berry, Pamela Mulhali

#### **Worksheet Rearranging Atoms:**

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