

19.3

STRENGTHS OF ACIDS AND BASES

Section Review

Objectives

1. Explain strong acids and bases (19.1)
2. Calculate K_a or K_b from percent ionization or K_b from K_a and vice versa (19.2)
3. Calculate the K_a or K_b of a weak acid or base from its pK_a or pK_b (19.2)
4. Calculate the pH of a weak acid or base solution (19.2)

Vocabulary

- | | |
|-----------------|-----------------|
| 1. strong acids | 2. strong bases |
| 3. weak acids | 4. K_b |
| 5. weak bases | 6. K_a |

Part A Completion

Fill in the spaces with the appropriate word or phrase. Write the chemical formula of the acid or base in the space provided. Do not use subscripts.

1. The strength of an acid or a base is determined by the _____ of the acid or base. The _____ of an acid is the concentration of _____ in a solution. The _____ of a base is the concentration of _____ in a solution. The _____ of an acid is the concentration of _____ in a solution. The _____ of a base is the concentration of _____ in a solution.
2. The strength of an acid or a base is determined by the _____ of the acid or base. The _____ of an acid is the concentration of _____ in a solution. The _____ of a base is the concentration of _____ in a solution. The _____ of an acid is the concentration of _____ in a solution. The _____ of a base is the concentration of _____ in a solution.
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