

Worksheet: Chemical Reaction Rates & Equilibrium

MULTIPLE CHOICE

- 1) Equilibrium is reached in a chemical reaction when _____ 1) _____
A) The rates of the opposing reactions become equal
B) The reactants are completely consumed
C) The forward and reverse reactions stop
D) The concentrations of all reactants and products become equal
- 2) In which direction will the point of equilibrium shift when the concentration of iron(III) ion decreases in the following equilibrium? _____ 2) _____
$$\text{Fe}^{3+}(\text{aq}) + \text{SCN}^{-}(\text{aq}) \rightleftharpoons \text{FeSCN}^{2+}(\text{aq})$$

A) Shift to the right B) No shift C) Shift to the left
- 3) In which direction will the point of equilibrium shift when the pressure is increased in the following equilibrium? _____ 3) _____
$$\text{N}_2(\text{g}) + 3 \text{H}_2(\text{g}) \rightleftharpoons 2 \text{NH}_3(\text{g})$$

A) Shift to the right B) No shift C) Shift to the left
- 4) In which direction will the point of equilibrium shift when a catalyst is added to the following equilibrium system? _____ 4) _____
$$2 \text{NO}(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2 \text{NO}_2(\text{g}) + 1.13 \text{ kJ}$$

A) No shift B) Shift to the right C) Shift to the left
- 5) In the following equilibrium, _____ 5) _____
$$\text{NaCl(s)} \rightleftharpoons \text{Na}^{+}(\text{aq}) + \text{Cl}^{-}(\text{aq})$$

When chloride ion is added, the concentration of sodium ions will
A) decrease B) increase C) not change
- 6) Ammonia, an essential chemical for modern civilization, is produced from the forward reaction of the following equilibrium _____ 6) _____
$$\text{N}_2(\text{g}) + 3 \text{H}_2(\text{g}) \rightleftharpoons 2 \text{NH}_3(\text{g}) + 92.5 \text{ kJ}$$

From the equilibrium, which of the following would favor the production of ammonia?
A) Reducing the concentration of ammonia
B) Increasing the pressure
C) Increasing the concentration of nitrogen gas
D) All of the above
- 7) Which one of the following factors would affect the rate of the chemical reaction? _____ 7) _____
A) Temperature B) Concentration of reactants
C) Catalyst D) All of the above

Workbook Answers Reaction Rates And Equilibrium

DIWAKAR EDUCATION HUB

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