

Consider the following equation:



- 1) What type of chemical reaction is taking place? _____
- 2) How many milliliters of 0.225 M HBr would be needed to react completely with 3.26 grams of magnesium hydroxide?
- 3) If 31.6 grams of magnesium hydroxide is combined with 68.0 mL of 0.725 M HBr, which is the limiting reagent? How many grams of magnesium bromide would be formed?
- 4) How many grams of the excess reagent will be left over after the reaction in part 3 is complete?

Work For Section 3 Stoichiometry

G Psacharopoulos



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