

Chapter # 04

→ Chemical Bonding

1. **Atoms combine to form molecules because:**
A. ✓ They have tendency to complete their outer most shell
B. They have no tendency to complete their outer most shell
C. They avoid inert gas structure
D. They tend to lose their stable structure
2. **Bond energy is expressed nowadays in:**
A. K Cals / mole
B. ✓ KJ mol^{-1}
C. A
D. Debye (μ)
3. **A molecule of O_2 is characterized of possessing:**
A. A single bond between two oxygen atoms
B. ✓ A double bond between two oxygen atoms
C. A triple bond between two oxygen atoms
D. None of the above
4. **The bond which is formed due to electrostatic forces of attraction between and negative ions is called:**
A. ✓ Ionic bond
B. Covalent bond
C. Co-ordinate covalent bond
D. Polar bond
5. **If an element of group IA is combining with another element of group VIIA, open predict the nature of bonding:**
A. ✓ Electrovalent
B. Covalent
C. Dative
D. Polar covalent
6. **In HCl, the bond formed is:**
A. Ionic
B. Non polar covalent
C. ✓ Polar covalent bond
D. Co-ordinate covalent
7. **Ionic bonds are also called:**
A. Electrovalent bond
B. ✓ Electrovalent as well as electrostatic bond
C. Polar covalent bond
D. Dative bond

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Table of Contents Wordwise Chapter 6 Chemical Bond

1. Understanding the eBook Wordwise Chapter 6 Chemical Bond
 - The Rise of Digital Reading Wordwise Chapter 6 Chemical Bond
 - Advantages of eBooks Over Traditional Books
2. Identifying Wordwise Chapter 6 Chemical Bond
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Wordwise Chapter 6 Chemical Bond
 - User-Friendly Interface
4. Exploring eBook Recommendations from Wordwise Chapter 6 Chemical Bond
 - Personalized Recommendations

- Wordwise Chapter 6 Chemical Bond User Reviews and Ratings
- Wordwise Chapter 6 Chemical Bond and Bestseller Lists
- 5. Accessing Wordwise Chapter 6 Chemical Bond Free and Paid eBooks
 - Wordwise Chapter 6 Chemical Bond Public Domain eBooks
 - Wordwise Chapter 6 Chemical Bond eBook Subscription Services
 - Wordwise Chapter 6 Chemical Bond Budget-Friendly Options
- 6. Navigating Wordwise Chapter 6 Chemical Bond eBook Formats
 - ePub, PDF, MOBI, and More
 - Wordwise Chapter 6 Chemical Bond Compatibility with Devices
 - Wordwise Chapter 6 Chemical Bond Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Wordwise Chapter 6 Chemical Bond
 - Highlighting and Note-Taking Wordwise Chapter 6 Chemical Bond
 - Interactive Elements Wordwise Chapter 6 Chemical Bond
- 8. Staying Engaged with Wordwise Chapter 6 Chemical Bond
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Wordwise Chapter 6 Chemical Bond
- 9. Balancing eBooks and Physical Books Wordwise Chapter 6 Chemical Bond
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Wordwise Chapter 6 Chemical Bond
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Wordwise Chapter 6 Chemical Bond
 - Setting Reading Goals Wordwise Chapter 6 Chemical Bond
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Wordwise Chapter 6 Chemical Bond
 - Fact-Checking eBook Content of Wordwise Chapter 6 Chemical Bond

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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