

Chemistry Worksheets Class 10 on Chapter 1 Chemical Reactions and Equations - Set 2

Q-1: Balance the following chemical equations:
i) CS₂ + O₂ → CO₂ + SO₂
ii) SO₂ + H₂S → H₂O+ S
iii) CaC ₂ + H ₂ O → Ca(OH) ₂ + C ₂ H ₂
iv) Na + H ₂ O → NaOH + H ₂
v) HgO+ heat → Hg + O₁
Q-2: Fill in the blanks
i) Respiration involves the oxidation of in the body.
ii) When water is electrolytically decomposed, H ₂ and O ₂ gas are obtained at and
respectively.
iii) When H ₂ gas is passed over heated copper(II) oxide, the residue left behind is that of
iv) Aluminium is highly to corrosion.
v) The reactions that involve absorption of heat are called
Q-3: Salt formation occurs when small amount of acid is added to
i) water
ii) acid
iii) base
iv) pure water
Q-4: A visually impaired student can use which of the following as an acid-base indicator?
i) Litmus
ii) Red cabbage
iii) Onion
iv) Petunia Leaves

Q-5: Match the column I with appropriate entries of Column II

Column I	Column II	
A) Silver nitrate + Sodium chloride	i) Red	
B) Lead nitrate + Potassium Iodide	ii) Black	
C) Sodium sulphate + Barium chloride	iii) Yellow	

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Padhraic Smyth

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Table of Contents Word Wise Chemistry Chapter Two Review

- 1. Understanding the eBook Word Wise Chemistry Chapter Two Review
 - The Rise of Digital Reading Word Wise Chemistry Chapter Two Review
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Word Wise Chemistry Chapter Two Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Word Wise Chemistry Chapter Two Review
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Word Wise Chemistry Chapter Two Review
 - Personalized Recommendations
 - Word Wise Chemistry Chapter Two Review User Reviews and Ratings
 - Word Wise Chemistry Chapter Two Review and Bestseller Lists

- 5. Accessing Word Wise Chemistry Chapter Two Review Free and Paid eBooks
 - Word Wise Chemistry Chapter Two Review Public Domain eBooks
 - Word Wise Chemistry Chapter Two Review eBook Subscription Services
 - Word Wise Chemistry Chapter Two Review Budget-Friendly Options
- 6. Navigating Word Wise Chemistry Chapter Two Review eBook Formats
 - o ePub, PDF, MOBI, and More
 - Word Wise Chemistry Chapter Two Review Compatibility with Devices
 - Word Wise Chemistry Chapter Two Review Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Word Wise Chemistry Chapter Two Review
 - Highlighting and Note-Taking Word Wise Chemistry Chapter Two Review
 - Interactive Elements Word Wise Chemistry Chapter Two Review
- 8. Staying Engaged with Word Wise Chemistry Chapter Two Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Word Wise Chemistry Chapter Two Review
- 9. Balancing eBooks and Physical Books Word Wise Chemistry Chapter Two Review
 - ∘ Benefits of a Digital Library
 - Creating a Diverse Reading Collection Word Wise Chemistry Chapter Two Review
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Word Wise Chemistry Chapter Two Review
 - Setting Reading Goals Word Wise Chemistry Chapter Two Review
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Word Wise Chemistry Chapter Two Review
 - Fact-Checking eBook Content of Word Wise Chemistry Chapter Two Review
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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