

Name : _____

Writing Balanced Chemical Equations

Write balanced chemical equations for the following reactions:

- (a) aluminum bromide + chlorine $\rightarrow$ aluminum chloride + bromine
- (b) calcium sulfite $\rightarrow$ calcium oxide + sulfur dioxide
- (c) phosphoric acid + calcium hydroxide $\rightarrow$ calcium phosphate + water
- (d) silver nitrate + ammonium carbonate $\rightarrow$ silver carbonate + ammonium nitrate
- (e) lithium hydrogen carbonate $\rightarrow$ lithium carbonate + water + carbon dioxide
- (f) magnesium + silver acetate $\rightarrow$ magnesium acetate + silver
- (g) aluminum bromide + potassium sulfate $\rightarrow$ potassium bromide + aluminum sulfate
- (h) sodium chloride + fluorine $\rightarrow$ sodium fluoride + chlorine
- (i) copper + sulfuric acid $\rightarrow$ copper sulfate + water + oxygen
- (j) zinc + iron (III) chloride $\rightarrow$ zinc chloride + iron

Word Chemical Equations Answer Sheet

Michael Hoelscher

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Chemical Misconceptions Keith Taber, 2002 Part 2 provides strategies for dealing with some of the misconceptions that students have by including ready to use classroom resources **Chemistry Equations & Answers** Speedy

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Chemical Reactions, Materials and Particles Gill Murphy, 2006-03 Specials are a teaching resource designed for KS3 students whose literacy skills are considerably lower than their age These books have an older format to counteract the simple text and cover various topics They include activities visuals and assessment sheets as well as teacher pages ,

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