Word Equations

Write the skeleton equations for each of the following chemical reactions and then balance them:

 When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made.

$$BeCl_{2(aq)} + 2 AgNO_{3(aq)} \rightarrow Be(NO_3)_{2(aq)} + 2 AgCl_{(s)}$$

 When isopropanol (C₃H₈O) burns in oxygen, carbon dioxide, water, and heat are produced.

$$2 C_3H_8O_{(1)} + 9 O_{2(9)} \rightarrow 6 CO_{2(9)} + 8 H_2O_{(9)}$$

When dissolved sodium hydroxide reacts with sulfuric acid, aqueous sodium sulfate, water, and heat are formed.

4) When fluorine gas is put into contact with calcium metal at high temperatures, calcium fluoride powder is created in an exothermic reaction.

$$F_{2(g)} + Ca_{(s)} \rightarrow CaF_{2(s)}$$

When sodium metal reacts with iron (II) chloride, iron metal and sodium chloride are formed.

2 Na_(s) + FeCl_{2(s)}
$$\rightarrow$$
 2 NaCl_(s) + Fe_(s)

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- 1. Understanding the eBook Word Equations Answer Key Chemfiesta
 - The Rise of Digital Reading Word Equations Answer Key Chemfiesta
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Word Equations Answer Key Chemfiesta
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Word Equations Answer Key Chemfiesta
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Word Equations Answer Key Chemfiesta
 - Personalized Recommendations
 - Word Equations Answer Key Chemfiesta User Reviews and Ratings
 - Word Equations Answer Key Chemfiesta and Bestseller Lists
- 5. Accessing Word Equations Answer Key Chemfiesta Free and Paid eBooks
 - Word Equations Answer Key Chemfiesta Public Domain eBooks
 - Word Equations Answer Key Chemfiesta eBook Subscription Services
 - Word Equations Answer Key Chemfiesta Budget-Friendly Options
- 6. Navigating Word Equations Answer Key Chemfiesta eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Word Equations Answer Key Chemfiesta Compatibility with Devices
 - Word Equations Answer Key Chemfiesta Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Word Equations Answer Key Chemfiesta
 - Highlighting and Note-Taking Word Equations Answer Key Chemfiesta
 - Interactive Elements Word Equations Answer Key Chemfiesta
- 8. Staying Engaged with Word Equations Answer Key Chemfiesta
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Word Equations Answer Key Chemfiesta
- 9. Balancing eBooks and Physical Books Word Equations Answer Key Chemfiesta
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Word Equations Answer Key Chemfiesta
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Word Equations Answer Key Chemfiesta
 - Setting Reading Goals Word Equations Answer Key Chemfiesta
 - Carving Out Dedicated Reading Time

- 12. Sourcing Reliable Information of Word Equations Answer Key Chemfiesta
 - Fact-Checking eBook Content of Word Equations Answer Key Chemfiesta
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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