

1. Write your answer.

Math Skills

Determining pH

Below are eight sample problems and solutions. Write your answers to the corresponding problems on a separate sheet of paper. Write your answers to the corresponding problems.

Problem 1a

Below is the pH of a 0.1 M solution of the strong acid HCl. Determine the concentration.

Given/Find

Step 1: Write the given and unknown values.

Given: concentration of HCl is 0.1 M (or 0.1 mol/L) HCl

Find: pH

Step 2: Determine the concentration of hydronium ions.

concentration of HCl is 0.1 M (or 0.1 mol/L) HCl

HCl is a strong acid, so the concentration of hydronium ions is the same as the concentration of HCl.

concentration of H₃O⁺ ions is concentration of HCl (or 0.1 M) HCl = 0.1 M (or 0.1 mol/L)

Step 3: Calculate the H₃O⁺ concentration pH

pH = the negative of the power of ten used to describe the concentration of H₃O⁺ ions

concentration of H₃O⁺ ions = 0.1 M (or 10⁻¹ M)

pH = -log(10⁻¹)

Problem 1b

1. A strong acid, HCl, has a concentration of 0.1 M. Determine the pH of this solution.

2. A weak acid, HCl, has a concentration of 0.1 M. Determine the pH of this solution.

Skills Math Skills Determining Ph

**Peter J. Mikulecky,Michelle Rose
Gilman,Kate Brutlag**



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Research in Education ,1974

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