

Net Ionic Equations Worksheet

Write balanced chemical equations for dissolving the following compounds in water:

1. Solid sodium chloride



2. Solid barium nitrate



3. Solid magnesium hydroxide



4. Gaseous hydrochloric acid



5. Solid calcium phosphate



6. Solid iron (III) sulfide



7. Gaseous carbon dioxide (*Hint: Carbon dioxide will form carbonic acid first then decomposes into ions*)



Write balanced chemical equations and then net ionic equations for the following reactions:

8. Aqueous sodium sulfate and aqueous barium chloride



9. Aqueous lithium bromide and aqueous silver nitrate



10. Aqueous lead II acetate and aqueous sodium carbonate



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Lifelique



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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 **Net Ionic Equations** Lifeliqe, 2019 This lesson plan covers the various ways to depict chemical equations using net ionic equations **Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance)** Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these chemical reactions *Chemical Equations: Chemistry Essentials Practice Workbook with 200+ Reactions to Balance* Jake Mariska, 2021-05-08 A chemical equation is something you will encounter every day in chemistry It s a written representation using numbers and symbols of the process that occurs during a chemical reaction Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas

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