#### SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following guestions.

- What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water
- Define solubility. A measure of how much solute can dissolve in a given amount of solvent
- According to the graph, the solubility of any substance changes as temperature. changes.
- List the substances whose solubility decreases as temperature increases. NH and Ce2(5O4)
- Which substance is least affected by temperature changes? NaCl.
- How many grams of ammonium chloride (NH<sub>a</sub>CI) at 50°C? 50g
- NaCl and KCIO<sub>3</sub> have the same solubility at approximately 78°C.
- Which compound is least soluble in water at 10°C? KCIO<sub>1</sub>
- How many grams of KNO<sub>3</sub> can be dissolved at 50°C? 80g
- 10. Are the following solutions unsaturated, saturated, or supersaturated?
  - a. 45g of NaNO in 100 g of water at 30°C, saturated
  - b. 60g of KClO<sub>3</sub> in 100 g of water at 90°C, supersaturated
- 11. How many grams of sodium chloride, NaCl are required to saturate 100 grams of water at 100° C? 40q
- How many grams of NaNO<sub>3</sub> are required to saturate 100 grams of water at 75°C? 140g.
- How many grams of KI will saturate water at 20°C? 33g
- At what temperature would 25g of potassium chlorate (KCIO<sub>3</sub>) dissolve?
- At what temperature would 60g of NH<sub>4</sub>Cl dissolve? 70°C
- 89 q NaNO<sub>3</sub> is prepared at 30°C.
  - a) Will all of the salt dissolve? No
  - b) What mass of NaNO<sub>3</sub> will dissolve at this temperature? 950
- 17. If 50 grams of NH<sub>4</sub>Cl is dissolved at 50°C, how many additional grams NH<sub>4</sub>Cl would be needed to make the solution saturated at 80°C? 15q
- At 50°C, how many grams of KNO<sub>3</sub> will dissolve? 80g
- At 70°C, how many grams of cerium (III) sulfate (Ce<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>) dissolve?
- Determine if each of the following is unsaturated, saturated, or supersaturated. f. 78g of NaNO at 10°C, saturated
- a. 55g of NH<sub>2</sub> at 20°C supersaturated
- b. 10g of Ce<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> at 10°C unsaturated
- c. 110g of KNO<sub>3</sub> at 60°C. supersaturated
- d. 65g of NH<sub>4</sub>Cl at 80°C. saturated
- e. 12g of NH<sub>1</sub> at 90°C, supersaturated CPHS Chemistry

g. 145g of NaNO<sub>1</sub> at 80°C, saturated

h. 35g of NaClat 100°C. unsaturated

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