

27

Initial $[\text{CrO}_4^{2-}]$ **0.0024 M**Initial $[\text{CrO}_4^{2-}]$ **0.0024 M**

Volume of 0.0024 M K_2CrO_4	Total volume	$[CrO_4^{2-}]$	Absorbance
1. 1.00 mL	100. mL		0.060
2. 5.00 mL	100. mL		0.299
3. 10.00 mL	100. mL		0.600
4. 15.00 mL	100. mL		0.840

Molar absorption coefficient for $[\text{CrO}_5]^{2-}$

Average molar absorption coefficient

Standard deviation (show calculations)

B. Determination of the Solubility-Product Constant

	Absorbance	$[\text{CrO}_4^{2-}]$	$[\text{Ag}^+]$	K_{sp} of Ag_2CrO_4
1.	0.720			
2.	0.750			
3.	0.780			

Solubility Product Lab Answers

Judd E. Hollander



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