

## SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

1. What are the customary units of solubility on solubility curves? Degress Celsius and grams of solute/100g of water
2. Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
3. According to the graph, the solubility of any substance changes as temperature changes.
4. List the substances whose solubility decreases as temperature increases.  $\text{NH}_3$  and  $\text{Ce}_2(\text{SO}_4)_3$
5. Which substance is least affected by temperature changes?  $\text{NaCl}$
6. How many grams of ammonium chloride ( $\text{NH}_4\text{Cl}$ ) at  $50^\circ\text{C}$ ? 50g
7.  $\text{NaCl}$  and  $\text{KClO}_3$  have the same solubility at approximately  $78^\circ\text{C}$ .
8. Which compound is least soluble in water at  $10^\circ\text{C}$ ?  $\text{KClO}_3$
9. How many grams of  $\text{KNO}_3$  can be dissolved at  $50^\circ\text{C}$ ? 80g
10. Are the following solutions unsaturated, saturated, or supersaturated?
  - a. 45g of  $\text{NaNO}_3$  in 100 g of water at  $30^\circ\text{C}$ . saturated
  - b. 60g of  $\text{KClO}_3$  in 100 g of water at  $90^\circ\text{C}$ . supersaturated
11. How many grams of sodium chloride,  $\text{NaCl}$  are required to saturate 100 grams of water at  $100^\circ\text{C}$ ? 40g
12. How many grams of  $\text{NaNO}_3$  are required to saturate 100 grams of water at  $75^\circ\text{C}$ ? 140g
13. How many grams of  $\text{KI}$  will saturate water at  $20^\circ\text{C}$ ? 33g
14. At what temperature would 25g of potassium chlorate ( $\text{KClO}_3$ ) dissolve?  $60^\circ\text{C}$
15. At what temperature would 60g of  $\text{NH}_4\text{Cl}$  dissolve?  $70^\circ\text{C}$
16. 89 g  $\text{NaNO}_3$  is prepared at  $30^\circ\text{C}$ .
  - a) Will all of the salt dissolve? No
  - b) What mass of  $\text{NaNO}_3$  will dissolve at this temperature? 95g
17. If 50 grams of  $\text{NH}_4\text{Cl}$  is dissolved at  $50^\circ\text{C}$ , how many additional grams  $\text{NH}_4\text{Cl}$  would be needed to make the solution saturated at  $80^\circ\text{C}$ ? 15g
18. At  $50^\circ\text{C}$ , how many grams of  $\text{KNO}_3$  will dissolve? 80g
19. At  $70^\circ\text{C}$ , how many grams of cerium (III) sulfate ( $\text{Ce}_2(\text{SO}_4)_3$ ) dissolve? 5g
20. Determine if each of the following is unsaturated, saturated, or supersaturated.
  - a. 55g of  $\text{NH}_3$  at  $20^\circ\text{C}$ . supersaturated
  - b. 10g of  $\text{Ce}_2(\text{SO}_4)_3$  at  $10^\circ\text{C}$ . unsaturated
  - c. 110g of  $\text{KNO}_3$  at  $60^\circ\text{C}$ . supersaturated
  - d. 65g of  $\text{NH}_4\text{Cl}$  at  $80^\circ\text{C}$ . saturated
  - e. 12g of  $\text{NH}_3$  at  $90^\circ\text{C}$ . supersaturated
  - f. 78g of  $\text{NaNO}_3$  at  $10^\circ\text{C}$ . saturated
  - g. 145g of  $\text{NaNO}_3$  at  $80^\circ\text{C}$ . saturated
  - h. 35g of  $\text{NaCl}$  at  $100^\circ\text{C}$ . unsaturated

# Solubility Practice Answer Key

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## **Solubility Practice Answer Key:**

## The Enigmatic Realm of **Solubility Practice Answer Key**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Solubility Practice Answer Key** a literary masterpiece penned by way of a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those who partake in its reading experience.

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### **Table of Contents Solubility Practice Answer Key**

1. Understanding the eBook Solubility Practice Answer Key
  - The Rise of Digital Reading Solubility Practice Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Solubility Practice Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a Solubility Practice Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Solubility Practice Answer Key
  - Personalized Recommendations
  - Solubility Practice Answer Key User Reviews and Ratings

- Solubility Practice Answer Key and Bestseller Lists
- 5. Accessing Solubility Practice Answer Key Free and Paid eBooks
  - Solubility Practice Answer Key Public Domain eBooks
  - Solubility Practice Answer Key eBook Subscription Services
  - Solubility Practice Answer Key Budget-Friendly Options
- 6. Navigating Solubility Practice Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Solubility Practice Answer Key Compatibility with Devices
  - Solubility Practice Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Solubility Practice Answer Key
  - Highlighting and Note-Taking Solubility Practice Answer Key
  - Interactive Elements Solubility Practice Answer Key
- 8. Staying Engaged with Solubility Practice Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Solubility Practice Answer Key
- 9. Balancing eBooks and Physical Books Solubility Practice Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Solubility Practice Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Solubility Practice Answer Key
  - Setting Reading Goals Solubility Practice Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solubility Practice Answer Key
  - Fact-Checking eBook Content of Solubility Practice Answer Key
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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