
Applied Petroleum Reservoir Engineering

3rd Edition Solution Manual

CHAPTER 2

2.1 using Equation 2.4: $pV = nRT$

$$(a) \ V = \frac{1(10.73)(60 + 460)}{14.7} = 379.6 \text{ SCF}$$

$$(b) \ V = \frac{1(10.73)(32 + 460)}{14.7} = 359.1 \text{ SCF}$$

$$(c) \ V = \frac{1(10.73)(80 + 460)}{14.7 + 10/16} = 378.1 \text{ SCF}$$

$$(d) \ V = \frac{1(10.73)(60 + 460)}{15.025} = 371.4 \text{ SCF}$$

2.2 (a) number of moles of methane = $10/16 = .625$ moles

number of moles of ethane = $20/30 = .667$ moles

total number of moles = $.625 + .667 = 1.292$ moles

(b) from Equation 2.4

$$p = \frac{1.292(10.73)(550)}{500} = 15.25 \text{ psia}$$

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A Loxley



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Table of Contents Solution Manual For Textbooks Reservoir Engineering

1. Understanding the eBook Solution Manual For Textbooks Reservoir Engineering
 - The Rise of Digital Reading Solution Manual For Textbooks Reservoir Engineering
 - Advantages of eBooks Over Traditional Books
2. Identifying Solution Manual For Textbooks Reservoir Engineering
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solution Manual For Textbooks Reservoir Engineering
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solution Manual For Textbooks Reservoir Engineering
 - Personalized Recommendations
 - Solution Manual For Textbooks Reservoir Engineering User Reviews and Ratings
 - Solution Manual For Textbooks Reservoir Engineering and Bestseller Lists
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 - Highlighting and Note-Taking Solution Manual For Textbooks Reservoir Engineering
 - Interactive Elements Solution Manual For Textbooks Reservoir Engineering
- 8. Staying Engaged with Solution Manual For Textbooks Reservoir Engineering
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solution Manual For Textbooks Reservoir Engineering
- 9. Balancing eBooks and Physical Books Solution Manual For Textbooks Reservoir Engineering
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solution Manual For Textbooks Reservoir Engineering
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Solution Manual For Textbooks Reservoir Engineering
 - Setting Reading Goals Solution Manual For Textbooks Reservoir Engineering
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solution Manual For Textbooks Reservoir Engineering
 - Fact-Checking eBook Content of Solution Manual For Textbooks Reservoir Engineering
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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