

Understanding Solubility

What is the quantity of solute in a saturated solution at a certain temperature?

Activity 1

Prepare two solid solvents: a potassium dichromate solution for copper through heat of heat loss and repeat this solid solvent, like potassium dichromate, with a completely different solvent. In addition, always quantify all solute solvents for potassium dichromate at the same solvent level, so that they should have the same amount of solute in each solvent. This is done by measuring the amount of solute in each solvent and comparing it with the amount of solute in the solvent.

After the solution is made, you will observe that the solute is dissolved in the solvent. The amount of solute in the solvent is the same as the amount of solute in the solvent. This is the amount of solute in the solvent.

Table 1: – Potassium Dichromate

The following table shows the results of the experiment. It is a table showing the amount of solute in a solvent at a certain temperature. The solute is potassium dichromate and the solvent is water. The amount of solute in the solvent is the same as the amount of solute in the solvent. The amount of solute in the solvent is the same as the amount of solute in the solvent. The amount of solute in the solvent is the same as the amount of solute in the solvent.

Solute	Experiment 1 in 100 g solvent		Experiment 2 in 100 g solvent		Experiment 3 in 100 g solvent	
	Amount of solute (g/mol/g solvent)	Amount of solute (g/mol/g solvent)	Amount of solute (g/mol/g solvent)	Amount of solute (g/mol/g solvent)	Amount of solute (g/mol/g solvent)	Amount of solute (g/mol/g solvent)
1	10.0	10.0	10.0	10.0	10.0	10.0
2	10.0	10.0	10.0	10.0	10.0	10.0
3	10.0	10.0	10.0	10.0	10.0	10.0
4	10.0	10.0	10.0	10.0	10.0	10.0
5	10.0	10.0	10.0	10.0	10.0	10.0
6	10.0	10.0	10.0	10.0	10.0	10.0
7	10.0	10.0	10.0	10.0	10.0	10.0
8	10.0	10.0	10.0	10.0	10.0	10.0
9	10.0	10.0	10.0	10.0	10.0	10.0
10	10.0	10.0	10.0	10.0	10.0	10.0

1. Solvent (the one that is dissolved) is potassium dichromate (g/mol/g solvent) (Table 1).

2. Solvent (the one that is dissolved) is potassium dichromate (g/mol/g solvent) (Table 1).

3. Solvent (the one that is dissolved) is potassium dichromate (g/mol/g solvent) (Table 1).

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Atherton Seidell



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