

Table F

Solubility Guidelines for Aqueous solutions

Ions That Form Soluble Compounds	Exceptions	Ions That Form Insoluble Compounds*	Exceptions
Group 1 ions (Li^+ , Na^+ , etc.)		carbonate (CO_3^{2-})	when combined with Group 1 ions or ammonium (NH_4^+)
ammonium (NH_4^+)		chromate (CrO_4^{2-})	when combined with Group 1 ions, Ca^{2+} , Mg^{2+} , or ammonium (NH_4^+)
nitrate (NO_3^-)		phosphate (PO_4^{3-})	when combined with Group 1 ions or ammonium (NH_4^+)
acetate ($\text{C}_2\text{H}_3\text{O}_2^-$ or CH_3COO^-)		sulfide (S^{2-})	when combined with Group 1 ions or ammonium (NH_4^+)
hydrogen carbonate (HCO_3^-)		hydroxide (OH^-)	when combined with Group 1 ions, Ca^{2+} , Ba^{2+} , Sr^{2+} , or ammonium (NH_4^+)
chlorate (ClO_3^-)			
halides (Cl^- , Br^- , I^-)	when combined with Ag^+ , Pb^{2+} , or Hg_2^{2+}		
sulfates (SO_4^{2-})	when combined with Ag^+ , Ca^{2+} , Sr^{2+} , Ba^{2+} , or Pb^{2+}		

*compounds having very low solubility in H_2O

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Atherton Seidell



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ChemDiscovery Lab Manual Olga I. Agapova, 2002 Ate Science Plus 2002 LV Red Holt Rinehart & Winston, 2001-02
The Science Teacher, 1992 Some issues are accompanied by a CD ROM on a selected topic **Illinois Chemistry Teacher**, 1992 **Handbook of Aqueous Solubility Data** Samuel H. Yalkowsky, Yan He, Parijat Jain, 2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of *Handbook of Aqueous Solubility Data* Samuel H. Yalkowsky, 2003-03-26 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until now this information has been scattered throughout the literature Containing over 16 000 solubility data points for more than 4 000 organic compounds Handbook of Aqueous **A Study of the Solubility of Certain Organic Substances in Aqueous Solutions** John Henry Saylor, 1930 **Solubilities of Inorganic and Organic Substances** Atherton Seidell, 1911 **Solubilities of Inorganic and Organic Compounds** Atherton Seidell, 1919 *Solubilities of Inorganic and Organic Substances* Atherton Seidell, 1919 *Solubilities of Inorganic and Organic Compounds* Atherton Seidell, 1928 Solubility Behavior of Pharmaceuticals in Aqueous Solutions Mohit Chandra Gupta, 1952 *The Experimental Determination of Solubilities* G. T. Hefter, R. P. T. Tomkins, 2003-08 Table of contents Solubility and Solubilization in Aqueous Media Samuel Hyman Yalkowsky, 1999 This book aims to provide the reader with a working knowledge of the various means of controlling the solubility or dissolution rate of a drug or other solute in an aqueous medium The book begins with the factors which govern solubility in general and then looks at aqueous solubility in particular including the properties of liquid mixtures and the thermodynamics of solutions formed from mixing two components The bulk of the book is then devoted to techniques for altering solubility and dissolution rate of organic compounds in aqueous media It discusses in detail the most commonly used solubility enhancers buffers cosolvents surfactants and complexants Each chapter is self contained and emphasizes the details for applying the techniques First Outlines of a Dictionary of Solubilities of Chemical Substances Frank Humphreys Storer, 1864 **Solubilities of inorganic and organic compounds** Atherton Seidell, William F. Linke, 1952 **Solubility Curves of Aqueous Or Aqueous Salt Solutions of Methyl Acetate-methanol and the Liquid-liquid Equilibrium of Methyl Acetate-methanol-aqueous Salt Solutions** A. Nakamura, 1968 **Solubilities of inorganic and organic substances** Atherton Seidell, 1911 Solutions and Solubilities Michael R. J. Dack, 1976

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