Unit - 1

Solid State

A solid is defined as the state of matter which has definite shape and definite volume.

* Characteritic properties of the Solid State ?-

- 1. It has stronger intermolecular forces as compared to the other states of matter (Liquid and Gases).
- 2. In solid constituent particles (ion, atom or molecules) are nearer to each other as compared to other state at matter.
- 3. In solid constituent particles cannot move from their position. They can just ascillate about their mean position
- 4. Solids are rigidin nature.
- 5. Solids are not compressible or least compressible as compared to other state of matter.
- 6. Solids have greater density than liquid and gases.
- * Classification of Solid Based On the nature of arrangement of Constituent Particles:-

Crystalline Solids

Amorphous Solids

Solid State Notes In Target Publication

B Lingard

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