## Unit 1: Fundamentals of Chemistry Exercise Questions

### **Exercise Multiple Choice Question Answers**

1.	Industrial chemistry deals with the manufacturing of compounds:			
	(a) in the laboratory		(b) on micro scale	
	(c) on commercial scale		(d) on economic scale	
2.	Which one of the following can be separated by physical means?			
	(a) mixture	(b) element	(c) compound	(d) radical
3.	The most abundant element occurring in the oceans is:			
	(a) oxygen	(b) hydrogen	(c) nitrogen	(d) silicon
4.	Which one of the following element is found in most abundance in the earth's crust?			
	(a) oxygen	(b) aluminum	(c) silicon	(d) iron
5.	The third abundant gas found in the atmosphere is:			
	(a) carbon monoxide	(b) oxygen.	(c) nitrogen	(d) argon
6.	One amu (atomic mass unit) is equivalent to:			
	(a) 1.66 × 10 <sup>-24</sup> mg	(b) 1.66 × 10 <sup>-24</sup> g	(c). $1.66 \times 10^{-23}$ g	(d) $1.66 \times 10^{-23}$ g
7.	All of the followings are tri-atomic molecule except:			
	(a) H <sub>2</sub>	(b) O <sub>3</sub>	(c) H <sub>2</sub> O	(d) CO <sub>2</sub>
8.	The mass of one molecule of water is:			
	(a) 18 amu	(b) 18 g	(c) 18 mg	(d) 18 kg
9.	The molar mass of H <sub>2</sub> S0 <sub>4</sub> is			
	(a) 98 g	(b) 48 amu	(c) 4.8 g	(d) 98 amu
10.	Molar mass is usually expressed in grams. Which one of the following is molar mass of O <sub>2</sub> in amu?			
	(a) 32 amu		(b) 53. 2 × 10 <sup>-24</sup> amu	
	(c) 1.92 ×10 <sup>-25</sup> amu		(d) 192.64 × 10 <sup>-25</sup> amu	
11.	How many numbers of moles are equivalent to 8 grams of CO <sub>2</sub> ?			
	(a) 0.15	(b) 0.18	(c) 0.21	(d) 0.24
12.	Which one of the following pairs has the same number of ions?			
	(a) 1 mole of NaCl and 1 mole of MgCl <sub>2</sub>			
	(b) 1/2 mole of NaCl and 1/2 mole of MgCl <sub>2</sub>			
	(c) 1/2 mole of NaCl and 1/3 mole of MgCl <sub>2</sub>			
	(d) 1/3 mole of NaCl and 1/2 mole of MgCl <sub>2</sub>			
13.	Which one of the following pairs has the same mass?			
	(a) 1 mole of CO and 1 mole of N <sub>2</sub> (b) 1 mole of CO and 1 mole of CO			
	(c) I mole of O2 and I mole of N2		(d) 1 mole of O <sub>2</sub> and 1 mole of CO <sub>2</sub>	

# **Solved Exercises Of 12th Chemistry**

**M** Walker

#### **Solved Exercises Of 12th Chemistry:**

**Problems and Problem Solving in Chemistry Education** Georgios Tsaparlis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner Educart CBSE Class 12 Chemistry Question Bank 2025-26 on new Syllabus 2026 (Includes Past Years Solved Questions) Educart, 2025-05-26 Book Structure Chapter wise coverage with practice Qs and Unit Test Worksheets How Good are Educart Question Banks Based on the NCERT rationalised syllabusBased on CBSE guidelines you study exactly what you need for exams Includes real life examples to make learning practical and relatable Case based and assertion reason questions for deeper understanding Covers previous board exam questions and those from the DIKSHA platform Includes detailed solutions for NCERT Exemplar questions to boost confidence Topper's Corner shares expert guidance to avoid common mistakes Why Choose this Book Most Recommended CBSE Reference Book for Chapter wise Study **Educart NCERT Exemplar Class 12 Mathematics 2025 Problems** Solutions (For 2025-26 Board Exam) Educart, 2025-04-16 Book Structure Theory Based Solutions High Order Thinking Questions Why is Educart NCERT Exemplar Good for Class 12 Boards Based on the NCERT Rationalised Syllabus covers only the most relevant and updated content Detailed Explanations for All NCERT Questions Step by step solutions for complete conceptual clarity Theory Smart Tricks Simplifies complex topics and enhances understanding Important Questions from Previous Years Papers DIKSHA Platform This provides exposure to commonly asked and high weightage questions Problem Solution Exemplar Offers detailed solutions to all NCERT Exemplar problems for effective practice Why choose this book The Educart NCERT Exemplar Class 12 Book is highly recommended by students for its structured approach to learning Whether you are aiming for board exams or competitive entrance tests this book is a reliable resource for success **Educart CBSE** Class 12 Economics Question Bank 2025-26 on new Syllabus 2026 (Includes Past Years Solved Questions)

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- 1. Understanding the eBook Solved Exercises Of 12th Chemistry
  - The Rise of Digital Reading Solved Exercises Of 12th Chemistry
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Solved Exercises Of 12th Chemistry
  - Exploring Different Genres
  - $\circ\,$  Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Solved Exercises Of 12th Chemistry
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Solved Exercises Of 12th Chemistry

- Personalized Recommendations
- Solved Exercises Of 12th Chemistry User Reviews and Ratings
- Solved Exercises Of 12th Chemistry and Bestseller Lists
- 5. Accessing Solved Exercises Of 12th Chemistry Free and Paid eBooks
  - Solved Exercises Of 12th Chemistry Public Domain eBooks
  - Solved Exercises Of 12th Chemistry eBook Subscription Services
  - Solved Exercises Of 12th Chemistry Budget-Friendly Options
- 6. Navigating Solved Exercises Of 12th Chemistry eBook Formats
  - o ePub, PDF, MOBI, and More
  - Solved Exercises Of 12th Chemistry Compatibility with Devices
  - Solved Exercises Of 12th Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Solved Exercises Of 12th Chemistry
  - Highlighting and Note-Taking Solved Exercises Of 12th Chemistry
  - Interactive Elements Solved Exercises Of 12th Chemistry
- 8. Staying Engaged with Solved Exercises Of 12th Chemistry
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Solved Exercises Of 12th Chemistry
- 9. Balancing eBooks and Physical Books Solved Exercises Of 12th Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Solved Exercises Of 12th Chemistry
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Solved Exercises Of 12th Chemistry
  - Setting Reading Goals Solved Exercises Of 12th Chemistry
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solved Exercises Of 12th Chemistry

- Fact-Checking eBook Content of Solved Exercises Of 12th Chemistry
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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