

2003 AP[®] CHEMISTRY FREE-RESPONSE QUESTIONS

Your responses to the rest of the questions in this part of the examination will be graded on the basis of the accuracy and relevance of the information cited. Explanations should be clear and well organized. Examples and equations may be included in your responses where appropriate. Specific answers are preferable to broad, diffuse responses.

Answer BOTH Question 5 below AND Question 6 printed on page 12. Both of these questions will be graded. The Section II score weighting for these questions is 30 percent (15 percent each).

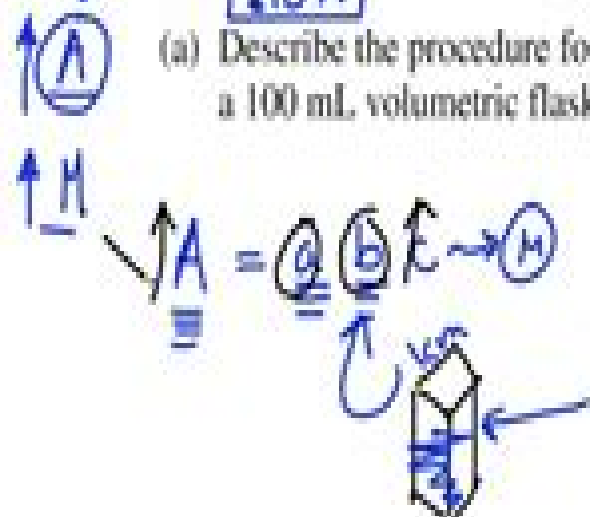
- 5 A student is instructed to determine the concentration of a solution of CoCl_2 based on absorption of light (spectrometric/colorimetric method). The student is provided with a 0.10 M solution of CoCl_2 with which to prepare standard solutions with concentrations of 0.020 M , 0.040 M , 0.060 M , and 0.080 M .

(a) Describe the procedure for diluting the 0.10 M solution to a concentration of 0.020 M using distilled water, a 100 mL volumetric flask, and a pipet or buret. Include specific amounts where appropriate.



$0.10\text{ M} \rightarrow \text{dilution } .02 \rightarrow .04 \rightarrow .06 \rightarrow .08$

$.10\text{ M} \rightarrow .02$



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Hons Chemistry and a Postgraduate Certificate in Education both from the University of Exeter in England In addition to writing this Crash Course Mr Dingle has written The Periodic Table Elements With Style How To Make A Universe With 92 Ingredients and SAT Chemistry Crash Course He is the 2011 winner of the School Library Association of the UK's Information Book Award and in 2012 was honored with the prestigious literary prize Wissenschaftsbuch des Jahre sponsored by the Austrian Ministry of Science and Research

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