#### Answers

 $Q = mc\Delta T$ , where Q = heat energy, m = mass, and  $\Delta T =$  change in temp. Remember,  $\Delta T = (T_{final} - T_{initial})$ . Show all work and proper units.

 A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = Q = 1086.75 = 0.46 \text{ J/g}^{\circ}\text{C}$$
  
m(Tr-Ti) 15.75(175-25)

 How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$Q = mC(T_1 - T_1) = 10.0g (0.90J/g^{\circ}C)(55-22) = 297 J$$

 Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = Q = \frac{67500 \text{ J}}{m(T_1-T_1)} = \frac{1.8 \text{ J/g}^{\circ}C}{(1500 \text{ g})(57-32)}$$

 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_r-T_i) = 100g(4.184J/g^{\circ}C)(37 - 4) = 14000 J$$

 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the specific heat capacity of mercury.

$$C = Q = 455 J = 0.14 J/g^{\circ}C$$
  
 $m(T_{i}-T_{i}) (25g)(155-25)$ 



# **Specific Heat Worksheet Answers With Work**

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comparisons and application notes for various types of fans blowers and compressors. The book also walks the reader through the general rules of thumb for vessels how cooling towers are sized based on parameters such as return temperature and supply temperature and specifications of refrigeration systems. Other chapters focus on pneumatic conveying blending and agitation energy conservation and process modeling. Online calculation tools Excel workbooks guidelines for hazardous materials and processes and a searchable Rules of Thumb library are included. Chemical engineers faced with fluid flow problems will find this book extremely useful Rules of Thumb for Chemical Engineers brings together solutions information and work arounds that engineers in the process industry need to get their job done. New material in the Fifth Edition includes physical properties for proprietary materials six new chapters including pharmaceutical biopharmaceutical sector heuristics process design with simulation software and guidelines for hazardous materials and processes. Now includes SI units throughout alongside imperial and now accompanied by online calculation tools and a searchable Rules of Thumb library.

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