

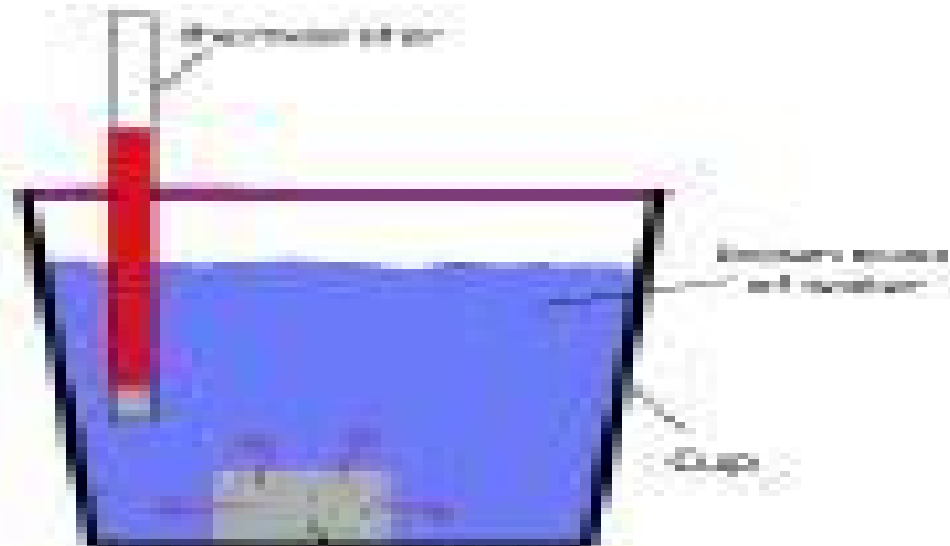
Specific Heat of Metal Lab

1 Purpose

To calculate the specific heat of an unknown metal.

2 Apparatus

- Unknown metal of mass m
- Clamp
- Weighing bottle
- Beaker
- Hot water
- Stirrer



3 PROCEDURE

1. Weigh out a substance by using a thermometer instead of a scale. Weigh it as to help us identify it.

Specific Heat Of A Metal Lab Answers

R Pring



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Scientific and Technical Aerospace Reports ,1995 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database **Lab Manual T/a Human Physiology** Bill W. Tillery,2001-03 *Liquid Metals, Liquid Metal Alloys and Their Applications* Defense Documentation Center (U.S.),1963 *Energy Research Abstracts* ,1993

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Abstract Bulletin Defense Documentation Center (U.S.),1961-04 **Subject Index to Unclassified ASTIA Documents**

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web answer the correct option is c explanation the molarity of a mixture m_{mix} can be calculated using the following formula $m_{\text{mix}} = \frac{m_1 + m_2 + m_3}{V_1 + V_2 + V_3}$ $1.5 \times 480 + 1.2 \times 520 + 480 + 520 + 720 + 624 = 1000$ 1.344 M q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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