

CHAPTER 2 WORKSHEET: CHEMISTRY OF LIFE

1. Chapter 2, Section 1: Nature of Matter

- a. Matter: anything that occupies space. Ex: Desk, Human, Air, ect.
- b. Atom: Smallest part of matter. It **can not** be broken down into smaller pieces. Composed of 3 particles: Protons (+), Electrons (-), and Neutrons (n)
- c. Element: Substance that is only made of one kind of atom. Ex: Hydrogen, Oxygen, ect.
 - i. C, H, N, O, P, S
- d. Compound: Two or more different atoms or elements joined together. Ex: Salt =
- e. Covalent Bond: Two or more atoms sharing electrons to form a molecule. Covalent bonds make molecules, ex: Water, Carbon Dioxide, and Oxygen Gas
- f. Hydrogen Bond: A weak chemical attraction between **polar** molecules. Ex: Two water molecules
- g. Ionic Bond: Gain or Loss of electrons that create a molecule
- h. Ion: An atom that has gained (-) electrons or (loss) electrons (+)

2. Chapter 2, Section 2: Water and Solutions

- a. How does water help maintain homeostasis?
 - i. Heats slowly
 - ii. Retains heat
 - iii. Carries Heat away through water vapor (sweating)
 - iv. Universal solvent (anything except for a polar substance can dissolve in it)
- b. What is the difference between adhesion & cohesion?
 - i. Adhesion: different substances
 - ii. Cohesion: alike substances
- c. What is the pH range for an Acid? Base?
 - i. Acid = 0 – 6.9
 - ii. Neutral = 7
 - iii. Base = 7.1 – 14
- d. What type of substance is an example of an acid? Of a base?
 - i. Acids: Lemon, Vinegar, Milk
 - ii. Bases: Household cleaning material

3. Chapter 2, Section 3: Chemistry of Cells

- a. Carbohydrates: Organic substances, made up of carbon, hydrogen and oxygen atoms. Always in the ratio of 1:2:1. Key source of energy.
 - i. They are made up of Monosaccharide: Single sugars (glucose)
 - ii. Disaccharides are double sugars: Sucrose
 - iii. Polysaccharides: Starch and cellulose

The Chemistry Of Life Chapter Answers

M Walker



The Chemistry Of Life Chapter Answers:

The Chemistry Of Life Chapter Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**The Chemistry Of Life Chapter Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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